Calcium sulfate

Other names: calcium sulfate dihydrate

gypsum

InChl=1S/Ca.H2O4S.2H2O/c;1-5(2,3)4;;/h;(H2,1,2,3,4);2*1H2/q+2;;;/p-2

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Formula: CaH4O6S

SMILES: O.O.O=S1(=O)O[Ca]O1

Mol. weight [g/mol]: 172.17

Sources

Effects of Na, Ca, Mg, and Al Chloride Salts on Dissolution and Phase Blacking Deposition Sulfate Dihydrate in Aqueous Solutions at 278.15 K to Massyrement and modelling of solubility for calcium sulfate dihydrate and calcium hydroxide in NaOH/KOH solutions:

https://www.doi.org/10.1021/acs.jced.5b00005

http://webbook.nist.gov/cgi/cbook.cgi?ID=B6004671&Units=SI

https://www.doi.org/10.1016/j.fluid.2010.06.012

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https://www.chemeo.com/cid/59-561-4/Calcium-sulfate.pdf

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