

# 4-Penten-1-ol, trichloroacetate

<b>Inchi:</b>	InChI=1S/C7H9Cl3O2/c1-2-3-4-5-12-6(11)7(8,9)10/h2H,1,3-5H2
<b>InchiKey:</b>	WMVZSMAZWDFZPH-UHFFFAOYSA-N
<b>Formula:</b>	C7H9Cl3O2
<b>SMILES:</b>	C=CCCCOC(=O)C(Cl)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	231.50

## Physical Properties

Property code	Value	Unit	Source
gf	-170.97	kJ/mol	Joback Method
hf	-363.15	kJ/mol	Joback Method
hfus	20.57	kJ/mol	Joback Method
hvap	51.52	kJ/mol	Joback Method
log10ws	-3.03		Crippen Method
logp	2.866		Crippen Method
mcvol	149.350	ml/mol	McGowan Method
pc	2787.66	kPa	Joback Method
rinpola	1228.00		NIST Webbook
ripola	1637.00		NIST Webbook
tb	541.59	K	Joback Method
tc	750.57	K	Joback Method
tf	331.23	K	Joback Method
vc	0.569	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	299.82	J/molxK	541.59	Joback Method
cpg	309.59	J/molxK	576.42	Joback Method
cpg	318.70	J/molxK	611.25	Joback Method
cpg	327.20	J/molxK	646.08	Joback Method
cpg	335.11	J/molxK	680.91	Joback Method
cpg	342.47	J/molxK	715.74	Joback Method
cpg	349.30	J/molxK	750.57	Joback Method
dvisc	0.0027774	Paxs	331.23	Joback Method

dvisc	0.0015461	Paxs	366.29	Joback Method
dvisc	0.0009534	Paxs	401.35	Joback Method
dvisc	0.0006354	Paxs	436.41	Joback Method
dvisc	0.0004498	Paxs	471.47	Joback Method
dvisc	0.0003340	Paxs	506.53	Joback Method
dvisc	0.0002578	Paxs	541.59	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R26514&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R26514&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/59-564-1/4-Penten-1-ol-trichloroacetate.pdf>

Generated by Cheméo on 2024-04-26 10:06:31.290561311 +0000 UTC m=+16415240.211138626.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.