

# Cyclohexane, 1,1,3-trimethyl-

<b>Other names:</b>	1,1,3-TRIMETHYLCYCLOHEXANE CYCLOGERANIOLANE
<b>Inchi:</b>	InChI=1S/C9H18/c1-8-5-4-6-9(2,3)7-8/h8H,4-7H2,1-3H3
<b>InchiKey:</b>	PYOLJOJPIPCRDY-UHFFFAOYSA-N
<b>Formula:</b>	C9H18
<b>SMILES:</b>	CC1CCCC(C)(C)C1
<b>Mol. weight [g/mol]:</b>	126.24
<b>CAS:</b>	3073-66-3

## Physical Properties

Property code	Value	Unit	Source
chl	-5882.30	kJ/mol	NIST Webbook
gf	36.15	kJ/mol	Joback Method
hf	-179.87	kJ/mol	Joback Method
hfus	5.67	kJ/mol	Joback Method
hvap	34.60	kJ/mol	Joback Method
ie	9.39	eV	NIST Webbook
log10ws	-3.00		Crippen Method
logp	3.223		Crippen Method
mcvol	126.810	ml/mol	McGowan Method
pc	2875.03	kPa	Joback Method
rinpol	840.00		NIST Webbook
rinpol	832.00		NIST Webbook
rinpol	840.00		NIST Webbook
rinpol	838.00		NIST Webbook
rinpol	841.00		NIST Webbook
rinpol	835.00		NIST Webbook
rinpol	839.00		NIST Webbook
rinpol	837.80		NIST Webbook
rinpol	834.00		NIST Webbook
rinpol	804.00		NIST Webbook
rinpol	839.75		NIST Webbook
rinpol	845.40		NIST Webbook
rinpol	836.50		NIST Webbook
rinpol	836.80		NIST Webbook
rinpol	850.90		NIST Webbook
rinpol	851.00		NIST Webbook

rinpol	846.00		NIST Webbook
rinpol	846.30		NIST Webbook
rinpol	847.70		NIST Webbook
rinpol	847.70		NIST Webbook
rinpol	840.00		NIST Webbook
rinpol	846.00		NIST Webbook
rinpol	840.00		NIST Webbook
rinpol	841.00		NIST Webbook
rinpol	842.00		NIST Webbook
rinpol	846.00		NIST Webbook
rinpol	850.00		NIST Webbook
rinpol	836.00		NIST Webbook
rinpol	839.98		NIST Webbook
rinpol	845.00		NIST Webbook
rinpol	830.00		NIST Webbook
rinpol	856.00		NIST Webbook
rinpol	849.00		NIST Webbook
rinpol	837.80		NIST Webbook
rinpol	835.50		NIST Webbook
rinpol	832.28		NIST Webbook
rinpol	833.00		NIST Webbook
rinpol	830.00		NIST Webbook
rinpol	836.00		NIST Webbook
rinpol	833.00		NIST Webbook
tb	409.68 ± 0.30	K	NIST Webbook
tb	411.00 ± 2.00	K	NIST Webbook
tb	410.65 ± 1.50	K	NIST Webbook
tb	411.00 ± 2.00	K	NIST Webbook
tb	410.00 ± 2.00	K	NIST Webbook
tb	412.15 ± 1.50	K	NIST Webbook
tb	409.50 ± 0.20	K	NIST Webbook
tb	409.77 ± 0.03	K	NIST Webbook
tb	410.25 ± 1.00	K	NIST Webbook
tb	409.78 ± 0.30	K	NIST Webbook
tb	409.80	K	NIST Webbook
tb	409.78	K	KDB
tc	627.99	K	Joback Method
tf	204.55 ± 0.50	K	NIST Webbook
vc	0.469	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	255.95	J/mol×K	420.44	Joback Method
cpg	275.52	J/mol×K	455.03	Joback Method
cpg	293.86	J/mol×K	489.62	Joback Method
cpg	311.08	J/mol×K	524.21	Joback Method
cpg	327.25	J/mol×K	558.80	Joback Method
cpg	342.48	J/mol×K	593.40	Joback Method
cpg	356.86	J/mol×K	627.99	Joback Method
hvapt	37.70	kJ/mol	379.50	NIST Webbook
hvapt	38.60	kJ/mol	368.50	NIST Webbook
hvapt	38.40	kJ/mol	369.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.37902e+01
Coeff. B	-3.24337e+03
Coeff. C	-5.63770e+01
Temperature range (K), min.	296.58
Temperature range (K), max.	438.91

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	7.92667e+01
Coeff. B	-7.41233e+03
Coeff. C	-9.57566e+00
Coeff. D	6.21403e-06
Temperature range (K), min.	348.15
Temperature range (K), max.	411.15

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3073663&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3073663&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=569">https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=569</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=569">https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=569</a>

# Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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