

# Diethylene glycol, decyl ether, acetate

<b>Other names:</b>	2-(2-decyloxy-ethoxy)-ethanol, acetate
<b>Inchi:</b>	InChI=1S/C16H32O4/c1-3-4-5-6-7-8-9-10-11-18-12-13-19-14-15-20-16(2)17/h3-15H2,1-2
<b>InchiKey:</b>	ASRPERSRLFXPEG-UHFFFAOYSA-N
<b>Formula:</b>	C16H32O4
<b>SMILES:</b>	CCCCCCCCCOCCOCCOC(C)=O
<b>Mol. weight [g/mol]:</b>	288.42

## Physical Properties

Property code	Value	Unit	Source
gf	-360.08	kJ/mol	Joback Method
hf	-882.81	kJ/mol	Joback Method
hfus	42.36	kJ/mol	Joback Method
hvap	65.19	kJ/mol	Joback Method
log10ws	-3.56		Crippen Method
logp	3.723		Crippen Method
mvol	255.480	ml/mol	McGowan Method
pc	1330.04	kPa	Joback Method
rinpol	2007.10		NIST Webbook
tb	686.61	K	Joback Method
tc	855.74	K	Joback Method
tf	386.70	K	Joback Method
vc	0.992	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	737.63	J/molxK	686.61	Joback Method
cpg	755.14	J/molxK	714.80	Joback Method
cpg	771.89	J/molxK	742.99	Joback Method
cpg	787.86	J/molxK	771.17	Joback Method
cpg	803.07	J/molxK	799.36	Joback Method
cpg	817.52	J/molxK	827.55	Joback Method
cpg	831.20	J/molxK	855.74	Joback Method
dvisc	0.0011272	Paxs	386.70	Joback Method

dvisc	0.0005491	Paxs	436.69	Joback Method
dvisc	0.0003101	Paxs	486.67	Joback Method
dvisc	0.0001948	Paxs	536.65	Joback Method
dvisc	0.0001325	Paxs	586.64	Joback Method
dvisc	0.0000957	Paxs	636.62	Joback Method
dvisc	0.0000725	Paxs	686.61	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R184377&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R184377&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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