

# 1,3-Propanediol, 2-methyl-2-n-propyl-, dinitrate

<b>Other names:</b>	2-methyl-2-propylpropane-1,3-diyl dinitrate
<b>Inchi:</b>	InChI=1S/C7H14N2O6/c1-3-4-7(2,5-14-8(10)11)6-15-9(12)13/h3-6H2,1-2H3
<b>InchiKey:</b>	BLJBDLGFKNXUCB-UHFFFAOYSA-N
<b>Formula:</b>	C7H14N2O6
<b>SMILES:</b>	CCCC(C)(CO[N+](=O)[O-])CO[N+](=O)[O-]
<b>Mol. weight [g/mol]:</b>	222.20
<b>CAS:</b>	10605-24-0

## Physical Properties

Property code	Value	Unit	Source
gf	-128.00	kJ/mol	Joback Method
hf	-482.52	kJ/mol	Joback Method
hfus	31.57	kJ/mol	Joback Method
hvap	67.88	kJ/mol	Joback Method
log10ws	-2.83		Crippen Method
logp	1.209		Crippen Method
mcvol	156.070	ml/mol	McGowan Method
pc	2826.33	kPa	Joback Method
tb	704.85	K	Joback Method
tc	935.06	K	Joback Method
tf	502.75	K	Joback Method
vc	0.617	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	440.70	J/mol×K	704.85	Joback Method
cpg	452.67	J/mol×K	743.22	Joback Method
cpg	463.77	J/mol×K	781.59	Joback Method
cpg	474.01	J/mol×K	819.95	Joback Method
cpg	483.42	J/mol×K	858.32	Joback Method
cpg	492.02	J/mol×K	896.69	Joback Method
cpg	499.85	J/mol×K	935.06	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10605240&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10605240&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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