

# 2,4-Dimethylthiophenol

<b>Other names:</b>	2,4-Dimethylbenzenethiol 2,4-Thioxylenol 2,4-Xylenethiol Benzenethiol, 2,4-dimethyl-
<b>Inchi:</b>	InChI=1S/C8H10S/c1-6-3-4-8(9)7(2)5-6/h3-5,9H,1-2H3
<b>InchiKey:</b>	AMNLXDDJGGTIPL-UHFFFAOYSA-N
<b>Formula:</b>	C8H10S
<b>SMILES:</b>	<chem>Cc1ccc(S)c(C)c1</chem>
<b>Mol. weight [g/mol]:</b>	138.23
<b>CAS:</b>	13616-82-5

## Physical Properties

Property code	Value	Unit	Source
gf	139.02	kJ/mol	Joback Method
hf	43.62	kJ/mol	Joback Method
hfus	13.78	kJ/mol	Joback Method
hvap	43.74	kJ/mol	Joback Method
log10ws	-2.94		Crippen Method
logp	2.592		Crippen Method
mcvol	116.170	ml/mol	McGowan Method
pc	3805.69	kPa	Joback Method
rinpol	1206.70		NIST Webbook
tb	481.94	K	Joback Method
tc	719.76	K	Joback Method
tf	267.84	K	Joback Method
vc	0.429	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	221.55	J/mol×K	481.94	Joback Method
cpg	233.86	J/mol×K	521.58	Joback Method
cpg	245.45	J/mol×K	561.21	Joback Method
cpg	256.34	J/mol×K	600.85	Joback Method

cpg	266.57	J/mol×K	640.49	Joback Method
cpg	276.15	J/mol×K	680.13	Joback Method
cpg	285.11	J/mol×K	719.76	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.41805e+01
Coeff. B	-3.93877e+03
Coeff. C	-7.49100e+01
Temperature range (K), min.	358.42
Temperature range (K), max.	519.02

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13616825&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13616825&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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