

# Benzeneethanol, 3-hydroxy-

<b>Other names:</b>	Phenethyl alcohol, m-hydroxy- m-Hydroxyphenethyl alcohol m-Phenethyl alcohol 3-(2-Hydroxyethyl)phenol 3-Hydroxybenzeneethanol 3-hydroxyphenethyl alcohol
<b>Inchi:</b>	InChI=1S/C8H10O2/c9-5-4-7-2-1-3-8(10)6-7/h1-3,6,9-10H,4-5H2
<b>InchiKey:</b>	AMQIPHZFLIDOCB-UHFFFAOYSA-N
<b>Formula:</b>	C8H10O2
<b>SMILES:</b>	OCCc1cccc(O)c1
<b>Mol. weight [g/mol]:</b>	138.16
<b>CAS:</b>	13398-94-2

## Physical Properties

Property code	Value	Unit	Source
gf	-162.55	kJ/mol	Joback Method
hf	-301.46	kJ/mol	Joback Method
hfus	20.39	kJ/mol	Joback Method
hvap	65.37	kJ/mol	Joback Method
log10ws	-1.09		Crippen Method
logp	0.927		Crippen Method
mcvol	111.560	ml/mol	McGowan Method
pc	5001.51	kPa	Joback Method
rinpol	1384.00		NIST Webbook
ripol	2318.00		NIST Webbook
ripol	2318.00		NIST Webbook
ripol	2415.00		NIST Webbook
tb	581.92	K	Joback Method
tc	791.22	K	Joback Method
tf	378.88	K	Joback Method
vc	0.360	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	269.14	J/mol×K	581.92	Joback Method
cpg	278.61	J/mol×K	616.80	Joback Method
cpg	287.43	J/mol×K	651.69	Joback Method
cpg	295.68	J/mol×K	686.57	Joback Method
cpg	303.42	J/mol×K	721.46	Joback Method
cpg	310.72	J/mol×K	756.34	Joback Method
cpg	317.63	J/mol×K	791.22	Joback Method
dvisc	0.0025798	Paxs	378.88	Joback Method
dvisc	0.0008141	Paxs	412.72	Joback Method
dvisc	0.0003060	Paxs	446.56	Joback Method
dvisc	0.0001320	Paxs	480.40	Joback Method
dvisc	0.0000636	Paxs	514.24	Joback Method
dvisc	0.0000335	Paxs	548.08	Joback Method
dvisc	0.0000191	Paxs	581.92	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	443.70	K	0.50	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13398942&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13398942&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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