

# Butane, 1,1,1,2,2,3,3-heptafluoro-4-iodo-

<b>Other names:</b>	Heptafluorobutyl iodide 2,2,3,3,4,4,4-Heptafluoro-1-iodobutane 1H,1H-Heptafluoro-1-iodobutane 1H,1H-Heptafluorobutyl iodide 1-Iodo-1H,1H-perfluorobutane 1,1,1,2,2,3,3-heptafluoro-4-iodobutane
<b>Inchi:</b>	InChI=1S/C4H2F7I/c5-2(6,1-12)3(7,8)4(9,10)11/h1H2
<b>InchiKey:</b>	AFHPVXVEXINDFS-UHFFFAOYSA-N
<b>Formula:</b>	C4H2F7I
<b>SMILES:</b>	FC(F)(F)C(F)(F)C(F)(F)CI
<b>Mol. weight [g/mol]:</b>	309.95
<b>CAS:</b>	374-98-1

## Physical Properties

Property code	Value	Unit	Source
gf	-1314.23	kJ/mol	Joback Method
hf	-1448.04	kJ/mol	Joback Method
hfus	9.84	kJ/mol	Joback Method
hvap	24.26	kJ/mol	Joback Method
ie	9.96 ± 0.02	eV	NIST Webbook
log10ws	-3.74		Crippen Method
logp	3.254		Crippen Method
mcvol	105.430	ml/mol	McGowan Method
pc	2823.32	kPa	Joback Method
tb	369.26	K	Joback Method
tc	536.77	K	Joback Method
tf	204.29	K	Joback Method
vc	0.441	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	199.37	J/mol×K	369.26	Joback Method
cpg	208.71	J/mol×K	397.18	Joback Method

cpg	217.27	J/mol×K	425.10	Joback Method
cpg	225.07	J/mol×K	453.02	Joback Method
cpg	232.16	J/mol×K	480.94	Joback Method
cpg	238.58	J/mol×K	508.85	Joback Method
cpg	244.38	J/mol×K	536.77	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	364.30 ± 0.30	K	98.70	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C374981&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C374981&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/59-818-9/Butane-1-1-1-2-2-3-3-heptafluoro-4-iodo.pdf>

Generated by Cheméo on 2024-04-29 13:03:22.850549344 +0000 UTC m=+16685051.771126660.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.