

Diethylmalonic acid, 4-chlorophenyl ethyl ester

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|----------------------|---|
| Inchi: | InChI=1S/C15H19ClO4/c1-4-15(5-2,13(17)19-6-3)14(18)20-12-9-7-11(16)8-10-12/h7-10H |
| InchiKey: | RMBSYOLXIBKWOZ-UHFFFAOYSA-N |
| Formula: | C15H19ClO4 |
| SMILES: | CCOC(=O)C(CC)(CC)C(=O)Oc1ccc(Cl)cc1 |
| Mol. weight [g/mol]: | 298.76 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -298.73 | kJ/mol | Joback Method |
| hf | -641.96 | kJ/mol | Joback Method |
| hfus | 30.62 | kJ/mol | Joback Method |
| hvap | 73.32 | kJ/mol | Joback Method |
| log10ws | -4.02 | | Crippen Method |
| logp | 3.615 | | Crippen Method |
| mcvol | 225.570 | ml/mol | McGowan Method |
| pc | 1944.08 | kPa | Joback Method |
| rinpol | 1857.00 | | NIST Webbook |
| rinpol | 1857.00 | | NIST Webbook |
| tb | 761.04 | K | Joback Method |
| tc | 977.37 | K | Joback Method |
| tf | 474.41 | K | Joback Method |
| vc | 0.854 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 622.21 | J/molxK | 761.04 | Joback Method |
| cpg | 636.18 | J/molxK | 797.09 | Joback Method |
| cpg | 649.11 | J/molxK | 833.15 | Joback Method |
| cpg | 661.04 | J/molxK | 869.20 | Joback Method |
| cpg | 671.99 | J/molxK | 905.26 | Joback Method |
| cpg | 682.01 | J/molxK | 941.31 | Joback Method |
| cpg | 691.13 | J/molxK | 977.37 | Joback Method |
| dvisc | 0.0007431 | Paxs | 474.41 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0004286 | Paxs | 522.18 | Joback Method |
| dvisc | 0.0002711 | Paxs | 569.95 | Joback Method |
| dvisc | 0.0001841 | Paxs | 617.73 | Joback Method |
| dvisc | 0.0001321 | Paxs | 665.50 | Joback Method |
| dvisc | 0.0000992 | Paxs | 713.27 | Joback Method |
| dvisc | 0.0000771 | Paxs | 761.04 | Joback Method |

Sources

| | |
|------------------------|---|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U369891&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |

Legend

| | |
|----------------------------|---|
| cp_g: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log₁₀ws: | Log ₁₀ of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| m_{cvol}: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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