

# 3-Butenoic acid, 4-phenyl-

<b>Other names:</b>	Styrylacetic acid 4-Phenyl-3-butenoic acid
<b>Inchi:</b>	InChI=1S/C10H10O2/c11-10(12)8-4-7-9-5-2-1-3-6-9/h1-7H,8H2,(H,11,12)/b7-4+
<b>InchiKey:</b>	PSCXFXNEYIHJST-QPJJXVBHSA-N
<b>Formula:</b>	C10H10O2
<b>SMILES:</b>	O=C(O)CC=Cc1ccccc1
<b>Mol. weight [g/mol]:</b>	162.19
<b>CAS:</b>	2243-53-0

## Physical Properties

Property code	Value	Unit	Source
gf	-39.79	kJ/mol	Joback Method
hf	-160.79	kJ/mol	Joback Method
hfus	21.59	kJ/mol	Joback Method
hvap	63.51	kJ/mol	Joback Method
log10ws	-2.23		Crippen Method
logp	2.174		Crippen Method
mcvol	131.140	ml/mol	McGowan Method
pc	3718.02	kPa	Joback Method
tb	605.09	K	Joback Method
tc	813.63	K	Joback Method
tf	360.15 ± 2.00	K	NIST Webbook
vc	0.492	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	306.84	J/mol×K	605.09	Joback Method
cpg	317.49	J/mol×K	639.85	Joback Method
cpg	327.42	J/mol×K	674.60	Joback Method
cpg	336.68	J/mol×K	709.36	Joback Method
cpg	345.31	J/mol×K	744.12	Joback Method
cpg	353.35	J/mol×K	778.87	Joback Method
cpg	360.85	J/mol×K	813.63	Joback Method

dvisc	0.0057762	Paxs	334.55	Joback Method
dvisc	0.0018366	Paxs	379.64	Joback Method
dvisc	0.0007448	Paxs	424.73	Joback Method
dvisc	0.0003592	Paxs	469.82	Joback Method
dvisc	0.0001968	Paxs	514.91	Joback Method
dvisc	0.0001188	Paxs	560.00	Joback Method
dvisc	0.0000773	Paxs	605.09	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2243530&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2243530&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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