

# Pyridine, 2,3,4,5-tetrahydro-

<b>Other names:</b>	«delta» 1-Piperideine 1-Piperideine 2,3,4,5-Tetrahydropyridine Tetrahydropyridine
<b>Inchi:</b>	InChI=1S/C5H9N/c1-2-4-6-5-3-1/h4H,1-3,5H2
<b>InchiKey:</b>	DWKUKQRKVC MOLP-UHFFFAOYSA-N
<b>Formula:</b>	C5H9N
<b>SMILES:</b>	C1=NCCCC1
<b>Mol. weight [g/mol]:</b>	83.13
<b>CAS:</b>	505-18-0

## Physical Properties

Property code	Value	Unit	Source
gf	170.12	kJ/mol	Joback Method
hf	56.88	kJ/mol	Joback Method
hfus	5.83	kJ/mol	Joback Method
hvap	33.96	kJ/mol	Joback Method
ie	8.95	eV	NIST Webbook
ie	8.20	eV	NIST Webbook
log10ws	-0.97		Crippen Method
logp	1.241		Crippen Method
mcvol	76.130	ml/mol	McGowan Method
pc	4938.44	kPa	Joback Method
rinpol	796.00		NIST Webbook
rinpol	796.00		NIST Webbook
tb	390.88	K	Joback Method
tc	615.07	K	Joback Method
tf	230.03	K	Joback Method
vc	0.284	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	134.36	J/molxK	390.88	Joback Method

cpg	148.44	J/mol×K	428.25	Joback Method
cpg	161.79	J/mol×K	465.61	Joback Method
cpg	174.42	J/mol×K	502.98	Joback Method
cpg	186.35	J/mol×K	540.34	Joback Method
cpg	197.59	J/mol×K	577.71	Joback Method
cpg	208.14	J/mol×K	615.07	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C505180&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C505180&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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