

6-Aminochrysene TFA

Inchi:

InChI=1S/C20H12F3NO/c21-20(22,23)19(25)24-18-11-17-13-6-2-1-5-12(13)9-10-15(17)

InchiKey:

LJGJMCASWROFLL-UHFFFAOYSA-N

Formula:

C20H12F3NO

SMILES:

O=C(Nc1cc2c3ccccc3ccc2c2ccccc12)C(F)(F)F

Mol. weight [g/mol]:

339.31

Physical Properties

Property code	Value	Unit	Source
gf	-100.13	kJ/mol	Joback Method
hf	-336.99	kJ/mol	Joback Method
hfus	40.01	kJ/mol	Joback Method
hvap	78.73	kJ/mol	Joback Method
log10ws	-7.75		Crippen Method
logp	5.647		Crippen Method
mcvol	227.380	ml/mol	McGowan Method
pc	2149.31	kPa	Joback Method
rinpol	443.06		NIST Webbook
tb	854.18	K	Joback Method
tc	1091.34	K	Joback Method
tf	584.02	K	Joback Method
vc	0.897	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	658.72	J/mol×K	854.18	Joback Method
cpg	670.47	J/mol×K	893.71	Joback Method
cpg	681.65	J/mol×K	933.23	Joback Method
cpg	692.43	J/mol×K	972.76	Joback Method
cpg	703.02	J/mol×K	1012.28	Joback Method
cpg	713.60	J/mol×K	1051.81	Joback Method
cpg	724.35	J/mol×K	1091.34	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R537494&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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