

# (E,Z)-1,2-Diethylenecyclopentane

<b>Inchi:</b>	InChI=1S/C9H14/c1-3-8-6-5-7-9(8)4-2/h3-4H,5-7H2,1-2H3/b8-3-,9-4+
<b>InchiKey:</b>	PSMXGHHIXPUKSP-SNXNNQDSSA-N
<b>Formula:</b>	C9H14
<b>SMILES:</b>	CC=C1CCCC1=CC
<b>Mol. weight [g/mol]:</b>	122.21

## Physical Properties

Property code	Value	Unit	Source
gf	160.08	kJ/mol	Joback Method
hf	3.79	kJ/mol	Joback Method
hfus	12.58	kJ/mol	Joback Method
hvap	37.77	kJ/mol	Joback Method
log10ws	-3.19		Crippen Method
logp	3.063		Crippen Method
mcvol	118.210	ml/mol	McGowan Method
pc	3049.04	kPa	Joback Method
rinpol	969.00		NIST Webbook
rinpol	969.00		NIST Webbook
tb	438.55	K	Joback Method
tc	644.95	K	Joback Method
tf	227.05	K	Joback Method
vc	0.448	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	229.85	J/molxK	438.55	Joback Method
cpg	296.15	J/molxK	610.55	Joback Method
cpg	284.39	J/molxK	576.15	Joback Method
cpg	271.92	J/molxK	541.75	Joback Method
cpg	258.70	J/molxK	507.35	Joback Method
cpg	244.69	J/molxK	472.95	Joback Method
cpg	307.24	J/molxK	644.95	Joback Method
dvisc	0.0002265	Paxs	438.55	Joback Method

dvisc	0.0002817	Paxs	403.30	Joback Method
dvisc	0.0003653	Paxs	368.05	Joback Method
dvisc	0.0005006	Paxs	332.80	Joback Method
dvisc	0.0007391	Paxs	297.55	Joback Method
dvisc	0.0012116	Paxs	262.30	Joback Method
dvisc	0.0023157	Paxs	227.05	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R288385&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R288385&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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