

# Benzoic acid, 2,3,6-trichloro-

<b>Other names:</b>	Benzac 1,281 HC 1281 T-2 Trysben TCB 2,3,6-Trichlorobenzoic acid 2,3,6-TBA 2,3,6-TCBA Acide trichlorobenzoique Benzabar Benzac-1281 Benzak Fen-all Kyselina 2,3,6-trichlorbenzoova NCI-C60242 2,3,6-Tba(the herbicide) 2,3,6-TCB TCBA Tribac 2,3,6-Trichlorbenzoesaure Trichlorobenzoic acid Trisben
<b>Inchi:</b>	InChI=1S/C7H3Cl3O2/c8-3-1-2-4(9)6(10)5(3)7(11)12/h1-2H,(H,11,12)
<b>InchiKey:</b>	XZIDTOHMJBOSOX-UHFFFAOYSA-N
<b>Formula:</b>	C7H3Cl3O2
<b>SMILES:</b>	O=C(O)c1c(Cl)ccc(Cl)c1Cl
<b>Mol. weight [g/mol]:</b>	225.46
<b>CAS:</b>	50-31-7

## Physical Properties

Property code	Value	Unit	Source
gf	-209.95	kJ/mol	Joback Method
hf	-297.72	kJ/mol	Joback Method
hfus	25.04	kJ/mol	Joback Method
hvap	72.02	kJ/mol	Joback Method
log10ws	-3.59		Crippen Method

logp	3.345		Crippen Method
mvol	129.890	ml/mol	McGowan Method
pc	4109.14	kPa	Joback Method
tb	659.52	K	Joback Method
tc	883.58	K	Joback Method
tf	402.70 ± 0.20	K	NIST Webbook
vc	0.491	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	271.85	J/mol×K	846.23	Joback Method
cpg	275.65	J/mol×K	883.58	Joback Method
cpg	247.03	J/mol×K	659.52	Joback Method
cpg	252.82	J/mol×K	696.86	Joback Method
cpg	258.18	J/mol×K	734.21	Joback Method
cpg	263.13	J/mol×K	771.55	Joback Method
cpg	267.68	J/mol×K	808.89	Joback Method
dvisc	0.0000877	Paxs	659.52	Joback Method
dvisc	0.0001192	Paxs	621.79	Joback Method
dvisc	0.0012269	Paxs	433.14	Joback Method
dvisc	0.0006628	Paxs	470.87	Joback Method
dvisc	0.0003923	Paxs	508.60	Joback Method
dvisc	0.0002496	Paxs	546.33	Joback Method
dvisc	0.0001684	Paxs	584.06	Joback Method
hfust	23.85	kJ/mol	402.70	NIST Webbook
hfust	23.85	kJ/mol	402.70	NIST Webbook

## Sources

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C50317&Units=SI>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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