

# Cyclohexanone, 2-cyclohexylidene-

<b>Other names:</b>	Bicyclohexyliden-2-one Dianon 2-Cyclohexylidenecyclohexanone
<b>Inchi:</b>	InChI=1S/C12H18O/c13-12-9-5-4-8-11(12)10-6-2-1-3-7-10/h1-9H2
<b>InchiKey:</b>	TYDSIOSLHQWFOU-UHFFFAOYSA-N
<b>Formula:</b>	C12H18O
<b>SMILES:</b>	O=C1CCCCC1=C1CCCCC1
<b>Mol. weight [g/mol]:</b>	178.27
<b>CAS:</b>	1011-12-7

## Physical Properties

Property code	Value	Unit	Source
chs	-6987.70 ± 1.20	kJ/mol	NIST Webbook
gf	2.59	kJ/mol	Joback Method
hf	-213.80 ± 2.40	kJ/mol	NIST Webbook
hfs	-307.00 ± 2.00	kJ/mol	NIST Webbook
hfus	8.32	kJ/mol	Joback Method
hvap	49.64	kJ/mol	Joback Method
log10ws	-3.77		Crippen Method
logp	3.390		Crippen Method
mcvol	155.490	ml/mol	McGowan Method
pc	2912.39	kPa	Joback Method
tb	599.34	K	Joback Method
tc	851.04	K	Joback Method
tf	342.26	K	Joback Method
vc	0.569	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	406.50	J/mol×K	599.34	Joback Method
cpg	428.31	J/mol×K	641.29	Joback Method
cpg	448.65	J/mol×K	683.24	Joback Method
cpg	467.54	J/mol×K	725.19	Joback Method

cpg	485.00	J/mol×K	767.14	Joback Method
cpg	501.06	J/mol×K	809.09	Joback Method
cpg	515.74	J/mol×K	851.04	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1011127&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1011127&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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