

Phthalic acid, butyl 2-(2-nitrophenyl)ethyl ester

Inchi:	InChI=1S/C20H21NO6/c1-2-3-13-26-19(22)16-9-5-6-10-17(16)20(23)27-14-12-15-8-4-7-
InchiKey:	MPEMNHVDPNTZMH-UHFFFAOYSA-N
Formula:	C20H21NO6
SMILES:	CCCCOC(=O)c1ccccc1C(=O)OCCc1ccccc1[N+](=O)[O-]
Mol. weight [g/mol]:	371.38

Physical Properties

Property code	Value	Unit	Source
gf	-109.21	kJ/mol	Joback Method
hf	-506.37	kJ/mol	Joback Method
hfus	51.79	kJ/mol	Joback Method
hvap	100.89	kJ/mol	Joback Method
log10ws	-5.90		Crippen Method
logp	3.951		Crippen Method
mvol	277.440	ml/mol	McGowan Method
pc	1733.22	kPa	Joback Method
rinpol	2848.00		NIST Webbook
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tb	1024.74	K	Joback Method
tc	1269.43	K	Joback Method
tf	680.97	K	Joback Method
vc	1.069	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	875.97	J/mol×K	1024.74	Joback Method
cpg	885.81	J/mol×K	1065.52	Joback Method
cpg	894.23	J/mol×K	1106.30	Joback Method
cpg	901.28	J/mol×K	1147.09	Joback Method
cpg	907.01	J/mol×K	1187.87	Joback Method
cpg	911.45	J/mol×K	1228.65	Joback Method
cpg	914.67	J/mol×K	1269.43	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U377994&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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