

Diethylmalonic acid, 4-cyanophenyl hexyl ester

Inchi:	InChI=1S/C20H27NO4/c1-4-7-8-9-14-24-18(22)20(5-2,6-3)19(23)25-17-12-10-16(15-21)
InchiKey:	RMPZVJHGGWXLOV-UHFFFAOYSA-N
Formula:	C20H27NO4
SMILES:	CCCCCOC(=O)C(CC)(CC)C(=O)Oc1ccc(C#N)cc1
Mol. weight [g/mol]:	345.43

Physical Properties

Property code	Value	Unit	Source
gf	-111.52	kJ/mol	Joback Method
hf	-564.54	kJ/mol	Joback Method
hfus	40.87	kJ/mol	Joback Method
hvap	90.55	kJ/mol	Joback Method
log10ws	-5.36		Crippen Method
logp	4.394		Crippen Method
mcvol	285.160	ml/mol	McGowan Method
pc	1335.88	kPa	Joback Method
rinsol	2409.00		NIST Webbook
tb	940.09	K	Joback Method
tc	1160.16	K	Joback Method
tf	565.83	K	Joback Method
vc	1.111	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	901.87	J/molxK	940.09	Joback Method
cpg	914.91	J/molxK	976.77	Joback Method
cpg	926.80	J/molxK	1013.45	Joback Method
cpg	937.59	J/molxK	1050.13	Joback Method
cpg	947.34	J/molxK	1086.80	Joback Method
cpg	956.10	J/molxK	1123.48	Joback Method
cpg	963.90	J/molxK	1160.16	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U369604&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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