

# Benzoic acid, 3-amino-2,5-dichloro-, methyl ester

<b>Other names:</b>	2,5-Dichloro-3-aminobenzoic acid, methyl ester 3-Amino-2,5-dichlorobenzoic acid, methyl ester Amchem 65-81-b Amiben, methyl ester Benzoic acid, 3-amino-2,5-dichloro-, methylated Chloramben methyl Chloramben, methyl ester Methyl 3-amino-2,5-dichlorobenzoate Methyl chloramben NSC 190368
<b>Inchi:</b>	InChI=1S/C8H7Cl2NO2/c1-13-8(12)5-2-4(9)3-6(11)7(5)10/h2-3H,11H2,1H3
<b>InchiKey:</b>	DTSSCQVCVYZGSI-UHFFFAOYSA-N
<b>Formula:</b>	C8H7Cl2NO2
<b>SMILES:</b>	<chem>COC(=O)c1cc(Cl)cc(N)c1Cl</chem>
<b>Mol. weight [g/mol]:</b>	220.05
<b>CAS:</b>	7286-84-2

## Physical Properties

Property code	Value	Unit	Source
gf	-91.33	kJ/mol	Joback Method
hf	-248.82	kJ/mol	Joback Method
hfus	25.73	kJ/mol	Joback Method
hvap	66.23	kJ/mol	Joback Method
log10ws	-3.26		Aqueous Solubility Prediction Method
logp	2.362		Crippen Method
mcvol	141.720	ml/mol	McGowan Method
pc	3581.35	kPa	Joback Method
tb	647.74	K	Joback Method
tc	887.37	K	Joback Method
tf	335.09 ± 0.50	K	NIST Webbook
vc	0.526	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	305.36	J/mol×K	647.74	Joback Method
cpg	314.51	J/mol×K	687.68	Joback Method
cpg	323.02	J/mol×K	727.62	Joback Method
cpg	330.92	J/mol×K	767.55	Joback Method
cpg	338.18	J/mol×K	807.49	Joback Method
cpg	344.83	J/mol×K	847.43	Joback Method
cpg	350.84	J/mol×K	887.37	Joback Method

## Sources

**Aqueous Solubility Prediction Method:** <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C7286842&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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