

Drimenal

Inchi:

InChI=1S/C15H24O/c1-11-6-7-13-14(2,3)8-5-9-15(13,4)12(11)10-16/h6,10,12-13H,5,7-9H

InchiKey:

TUWKAEMGKORGLZ-KCQAQPDRSA-N

Formula:

C15H24O

SMILES:

CC1=CCC2C(C)(C)CCCC2(C)C1C=O

Mol. weight [g/mol]:

220.35

Physical Properties

Property code	Value	Unit	Source
gf	42.93	kJ/mol	Joback Method
hf	-281.44	kJ/mol	Joback Method
hfus	15.14	kJ/mol	Joback Method
hvap	54.25	kJ/mol	Joback Method
log10ws	-4.06		Crippen Method
logp	3.984		Crippen Method
mcvol	197.760	ml/mol	McGowan Method
pc	2106.13	kPa	Joback Method
rinpol	1657.00		NIST Webbook
ripol	2045.00		NIST Webbook
ripol	2045.00		NIST Webbook
tb	617.10	K	Joback Method
tc	844.17	K	Joback Method
tf	375.21	K	Joback Method
vc	0.754	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	548.72	J/mol×K	617.10	Joback Method
cpg	570.40	J/mol×K	654.94	Joback Method
cpg	590.90	J/mol×K	692.79	Joback Method
cpg	610.45	J/mol×K	730.63	Joback Method
cpg	629.30	J/mol×K	768.48	Joback Method
cpg	647.72	J/mol×K	806.32	Joback Method
cpg	665.93	J/mol×K	844.17	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R440494&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
ripola:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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