Methyl helifolen-15-oate

Inchi: InChI=1S/C16H24O2/c1-14(2)12-6-5-11(13(17)18-4)16(12)9-7-15(14,3)8-10-16/h7,9,11-1
InchiKey: JECYVYPGCVYBMP-UHFFFAOYSA-N
Formula: C16H24O2
SMILES: COC(=O)C1CCC2C13C=CC(C)(CC3)C2(C)C
Mol. weight [g/mol]: 248.36

Physical Properties

Property code	Value	Unit	Source
gf	6.04	kJ/mol	Joback Method
hf	-349.47	kJ/mol	Joback Method
hfus	14.66	kJ/mol	Joback Method
hvap	56.67	kJ/mol	Joback Method
log10ws	-3.71		Crippen Method
logp	3.568		Crippen Method
mcvol	206.860	ml/mol	McGowan Method
pc	2137.41	kPa	Joback Method
rinpol	1688.00		NIST Webbook
ripol	2132.00		NIST Webbook
tb	661.07	K	Joback Method
tc	893.85	K	Joback Method
tf	453.00	K	Joback Method
vc	0.788	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	610.03	J/mol×K	661.07	Joback Method
cpg	630.94	J/mol×K	699.87	Joback Method
cpg	651.10	J/mol×K	738.66	Joback Method
cpg	670.92	J/mol×K	777.46	Joback Method
cpg	690.83	J/mol×K	816.26	Joback Method
cpg	711.25	J/mol×K	855.05	Joback Method
cpg	732.59	J/mol×K	893.85	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R503134&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume