

# Ethylphenylmalonic acid diethyl ester

**Other names:**

Diethyl ethylphenylmalonate  
Propanedioic acid, ethylphenyl-, diethyl ester  
Malonic acid, ethylphenyl-, diethyl ester

**Inchi:**

InChI=1S/C15H20O4/c1-4-15(13(16)18-5-2,14(17)19-6-3)12-10-8-7-9-11-12/h7-11H,4-6H

**InchiKey:**

PKRVDBARWFJWEB-UHFFFAOYSA-N

**Formula:**

C15H20O4

**SMILES:**

CCOC(=O)C(CC)(C(=O)OCC)c1ccccc1

**Mol. weight [g/mol]:**

264.32

**CAS:**

76-67-5

## Physical Properties

Property code	Value	Unit	Source
gf	-277.17	kJ/mol	Joback Method
hf	-614.75	kJ/mol	Joback Method
hfus	26.81	kJ/mol	Joback Method
hvap	68.28	kJ/mol	Joback Method
log10ws	-2.60		Crippen Method
logp	2.461		Crippen Method
mcvol	213.330	ml/mol	McGowan Method
pc	2036.39	kPa	Joback Method
tb	718.63	K	Joback Method
tc	930.98	K	Joback Method
tf	431.97	K	Joback Method
vc	0.804	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	596.90	J/mol×K	718.63	Joback Method
cpg	612.22	J/mol×K	754.02	Joback Method
cpg	626.46	J/mol×K	789.41	Joback Method
cpg	639.68	J/mol×K	824.81	Joback Method
cpg	651.90	J/mol×K	860.20	Joback Method
cpg	663.17	J/mol×K	895.59	Joback Method

cpg	673.51	J/mol×K	930.98	Joback Method
dvisc	0.0010981	Paxs	431.97	Joback Method
dvisc	0.0005847	Paxs	479.75	Joback Method
dvisc	0.0003490	Paxs	527.52	Joback Method
dvisc	0.0002270	Paxs	575.30	Joback Method
dvisc	0.0001577	Paxs	623.08	Joback Method
dvisc	0.0001154	Paxs	670.85	Joback Method
dvisc	0.0000880	Paxs	718.63	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C76675&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C76675&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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