

# Propanoic acid, 3-chloro, (Z)-3-hexenyl ester

Inchi:	InChI=1S/C9H15ClO2/c1-2-3-4-5-8-12-9(11)6-7-10/h3-4H,2,5-8H2,1H3/b4-3-
InchiKey:	RCTBGSWMRQJSLJ-ARJAWSKDSA-N
Formula:	C9H15ClO2
SMILES:	CCC=CCCOC(=O)CCCI
Mol. weight [g/mol]:	190.67

## Physical Properties

Property code	Value	Unit	Source
gf	-140.73	kJ/mol	Joback Method
hf	-372.41	kJ/mol	Joback Method
hfus	26.25	kJ/mol	Joback Method
hvap	49.13	kJ/mol	Joback Method
log10ws	-2.46		Crippen Method
logp	2.515		Crippen Method
mcvol	153.050	ml/mol	McGowan Method
pc	2448.32	kPa	Joback Method
rinpol	1289.00		NIST Webbook
rinpol	1298.00		NIST Webbook
rinpol	1295.00		NIST Webbook
rinpol	1276.00		NIST Webbook
rinpol	1279.00		NIST Webbook
rinpol	1279.00		NIST Webbook
ripol	1790.00		NIST Webbook
ripol	1780.00		NIST Webbook
ripol	1779.00		NIST Webbook
tb	523.20	K	Joback Method
tc	710.07	K	Joback Method
tf	288.19	K	Joback Method
vc	0.593	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	334.06	J/mol×K	523.20	Joback Method

cpg	346.42	J/molxK	554.34	Joback Method
cpg	358.22	J/molxK	585.49	Joback Method
cpg	369.46	J/molxK	616.63	Joback Method
cpg	380.17	J/molxK	647.78	Joback Method
cpg	390.36	J/molxK	678.92	Joback Method
cpg	400.04	J/molxK	710.07	Joback Method
dvisc	0.0027621	Paxs	288.19	Joback Method
dvisc	0.0013787	Paxs	327.36	Joback Method
dvisc	0.0007983	Paxs	366.53	Joback Method
dvisc	0.0005137	Paxs	405.70	Joback Method
dvisc	0.0003572	Paxs	444.86	Joback Method
dvisc	0.0002635	Paxs	484.03	Joback Method
dvisc	0.0002034	Paxs	523.20	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R296316&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R296316&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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