

# 2-Bromobenzyl alcohol, 1-methylpropyl ether

<b>Inchi:</b>	InChI=1S/C11H15BrO/c1-3-9(2)13-8-10-6-4-5-7-11(10)12/h4-7,9H,3,8H2,1-2H3
<b>InchiKey:</b>	GEABBXXCBUVYNO-UHFFFAOYSA-N
<b>Formula:</b>	C11H15BrO
<b>SMILES:</b>	CCC(C)OCc1ccccc1Br
<b>Mol. weight [g/mol]:</b>	243.14

## Physical Properties

Property code	Value	Unit	Source
gf	51.40	kJ/mol	Joback Method
hf	-156.48	kJ/mol	Joback Method
hfus	20.85	kJ/mol	Joback Method
hvap	51.48	kJ/mol	Joback Method
log10ws	-4.39		Crippen Method
logp	3.764		Crippen Method
mcvol	165.460	ml/mol	McGowan Method
pc	2775.92	kPa	Joback Method
rinpola	1471.00		NIST Webbook
tb	570.88	K	Joback Method
tc	791.16	K	Joback Method
tf	319.70	K	Joback Method
vc	0.618	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	368.86	J/molxK	570.88	Joback Method
cpg	383.64	J/molxK	607.59	Joback Method
cpg	397.55	J/molxK	644.31	Joback Method
cpg	410.62	J/molxK	681.02	Joback Method
cpg	422.89	J/molxK	717.73	Joback Method
cpg	434.38	J/molxK	754.45	Joback Method
cpg	445.12	J/molxK	791.16	Joback Method
dvisc	0.0019905	Paxs	319.70	Joback Method
dvisc	0.0010473	Paxs	361.56	Joback Method

dvisc	0.0006296	Paxs	403.43	Joback Method
dvisc	0.0004165	Paxs	445.29	Joback Method
dvisc	0.0002958	Paxs	487.15	Joback Method
dvisc	0.0002218	Paxs	529.02	Joback Method
dvisc	0.0001735	Paxs	570.88	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U378184&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U378184&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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