

# 2,2-Dimethylthirane

<b>Other names:</b>	Thiirane, 2,2-dimethyl-
<b>Inchi:</b>	InChI=1S/C4H8S/c1-4(2)3-5-4/h3H2,1-2H3
<b>InchiKey:</b>	HGJOFJDIHKHKAU-UHFFFAOYSA-N
<b>Formula:</b>	C4H8S
<b>SMILES:</b>	CC1(C)CS1
<b>Mol. weight [g/mol]:</b>	88.17
<b>CAS:</b>	3772-13-2

## Physical Properties

Property code	Value	Unit	Source
chl	-3295.00 ± 1.00	kJ/mol	NIST Webbook
gf	77.92	kJ/mol	Joback Method
hf	7.41	kJ/mol	Joback Method
hfl	-24.00 ± 1.00	kJ/mol	NIST Webbook
hfus	1.61	kJ/mol	Joback Method
hvap	29.07	kJ/mol	Joback Method
log10ws	-1.39		Crippen Method
logp	1.512		Crippen Method
mcvol	72.710	ml/mol	McGowan Method
pc	4869.76	kPa	Joback Method
rinpol	728.00		NIST Webbook
rinpol	692.00		NIST Webbook
rinpol	692.00		NIST Webbook
tb	345.73	K	Joback Method
tc	553.36	K	Joback Method
tf	260.13	K	Joback Method
vc	0.261	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	114.52	J/mol×K	345.73	Joback Method
cpg	125.43	J/mol×K	380.33	Joback Method
cpg	135.24	J/mol×K	414.94	Joback Method

cpg	144.05	J/mol×K	449.54	Joback Method
cpg	151.98	J/mol×K	484.15	Joback Method
cpg	159.14	J/mol×K	518.75	Joback Method
cpg	165.64	J/mol×K	553.36	Joback Method
hvapt	37.00	kJ/mol	373.00	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3772132&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3772132&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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