

Chloral

Other names:	2,2,2-Trichloroacetaldehyde 2,2,2-Trichloroethanal ANHYDROUS CHLORAL Acetaldehyde, 2,2,2-trichloro- Acetaldehyde, trichloro- CCI3CHO CHLOAL Cloralio Grasex Rcra waste number U034 Trichloroacetaldehyde Trichloroacetaldehyde Trichloroethanal
Inchi:	InChI=1S/C2HCl3O/c3-2(4,5)1-6/h1H
InchiKey:	HFFLGKNGCAIQMO-UHFFFAOYSA-N
Formula:	C2HCl3O
SMILES:	O=CC(Cl)(Cl)Cl
Mol. weight [g/mol]:	147.39
CAS:	75-87-6

Physical Properties

Property code	Value	Unit	Source
affp	722.30	kJ/mol	NIST Webbook
basg	690.50	kJ/mol	NIST Webbook
gf	-166.51	kJ/mol	Joback Method
hf	-226.16	kJ/mol	Joback Method
hfl	-218.10	kJ/mol	NIST Webbook
hfus	8.40	kJ/mol	Joback Method
hvap	41.00	kJ/mol	NIST Webbook
ie	10.88	eV	NIST Webbook
ie	10.90	eV	NIST Webbook
log10ws	-1.50		Crippen Method
logp	1.556		Crippen Method
mcvol	77.330	ml/mol	McGowan Method
pc	4903.92	kPa	Joback Method
rinpol	695.00		NIST Webbook
rinpol	695.00		NIST Webbook

rmpol	715.00		NIST Webbook
rmpol	715.00		NIST Webbook
tb	370.90 ± 0.50	K	NIST Webbook
tb	371.00	K	NIST Webbook
tb	369.55 ± 0.30	K	NIST Webbook
tc	618.25	K	Joback Method
tf	246.48	K	Joback Method
vc	0.300	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	125.96	J/mol×K	582.36	Joback Method
cpg	110.05	J/mol×K	402.88	Joback Method
cpg	114.03	J/mol×K	438.78	Joback Method
cpg	117.58	J/mol×K	474.67	Joback Method
cpg	120.73	J/mol×K	510.57	Joback Method
cpg	123.51	J/mol×K	546.46	Joback Method
cpg	128.09	J/mol×K	618.25	Joback Method
cpl	151.50	J/mol×K	298.00	NIST Webbook
dvisc	0.0005710	Paxs	402.88	Joback Method
dvisc	0.0055677	Paxs	246.48	Joback Method
dvisc	0.0031770	Paxs	272.55	Joback Method
dvisc	0.0019994	Paxs	298.61	Joback Method
dvisc	0.0013554	Paxs	324.68	Joback Method
dvisc	0.0009735	Paxs	350.75	Joback Method
dvisc	0.0007320	Paxs	376.81	Joback Method
hvapt	36.60	kJ/mol	303.00	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.53081e+01
Coeff. B	-3.79402e+03
Coeff. C	-1.60800e+01
Temperature range (K), min.	268.67

Temperature range (K), max.	395.61
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Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	8.45517e+01
Coeff. B	-7.35150e+03
Coeff. C	-1.03261e+01
Coeff. D	7.13919e-06
Temperature range (K), min.	216.00
Temperature range (K), max.	565.00

Sources

KDB Vapor Pressure Data:	https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1771
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemed.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
KDB:	https://www.thermo.com/files/research/kdb/mol/mol1771.mol
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C75876&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure

Legend

affp:	Proton affinity
basg:	Gas basicity
cpg:	Ideal gas heat capacity
cpl:	Liquid phase heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfl:	Liquid phase enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
ie:	Ionization energy

log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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