

Fumaric acid, isobutyl pentafluorophenyl ester

Inchi: InChI=1S/C14H11F5O4/c1-6(2)5-22-7(20)3-4-8(21)23-14-12(18)10(16)9(15)11(17)13(14)
InchiKey: CPCPEQPLOBNXGY-ONEGZZNKSA-N
Formula: C14H11F5O4
SMILES: CC(C)COC(=O)C=CC(=O)Oc1c(F)c(F)c(F)c(F)c1F
Mol. weight [g/mol]: 338.23

Physical Properties

Property code	Value	Unit	Source
gf	-1232.85	kJ/mol	Joback Method
hf	-1511.32	kJ/mol	Joback Method
hfus	41.76	kJ/mol	Joback Method
hvap	66.14	kJ/mol	Joback Method
log10ws	-4.43		Crippen Method
logp	3.043		Crippen Method
mcvol	203.790	ml/mol	McGowan Method
pc	1766.90	kPa	Joback Method
rinqol	1601.00		NIST Webbook
tb	723.95	K	Joback Method
tc	908.85	K	Joback Method
tf	463.75	K	Joback Method
vc	0.824	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	552.01	J/molxK	723.95	Joback Method
cpg	563.05	J/molxK	754.77	Joback Method
cpg	573.44	J/molxK	785.58	Joback Method
cpg	583.21	J/molxK	816.40	Joback Method
cpg	592.33	J/molxK	847.22	Joback Method
cpg	600.83	J/molxK	878.03	Joback Method
cpg	608.68	J/molxK	908.85	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U348095&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
m cvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/60-630-5/Fumaric-acid-isobutyl-pentafluorophenyl-ester.pdf>

Generated by Cheméo on 2024-04-20 05:44:55.565254573 +0000 UTC m=+15881144.485831888.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.