

Heptane, 2-bromo-

Other names:	2-Bromoheptane 2-Heptyl bromide
Inchi:	InChI=1S/C7H15Br/c1-3-4-5-6-7(2)8/h7H,3-6H2,1-2H3
InchiKey:	HLAUCEOFKXNF-UHFFFAOYSA-N
Formula:	C7H15Br
SMILES:	CCCCC(C)Br
Mol. weight [g/mol]:	179.10
CAS:	1974-04-5

Physical Properties

Property code	Value	Unit	Source
gf	19.94	kJ/mol	Joback Method
hf	-166.76	kJ/mol	Joback Method
hfus	15.65	kJ/mol	Joback Method
hvap	37.22	kJ/mol	Joback Method
log10ws	-3.30		Crippen Method
logp	3.350		Crippen Method
mvol	126.990	ml/mol	McGowan Method
pc	3042.32	kPa	Joback Method
rinpol	1037.00		NIST Webbook
rinpol	1047.00		NIST Webbook
rinpol	1026.00		NIST Webbook
rinpol	986.00		NIST Webbook
rinpol	1037.00		NIST Webbook
rinpol	986.00		NIST Webbook
ripol	1173.00		NIST Webbook
ripol	1173.00		NIST Webbook
ripol	1164.00		NIST Webbook
ripol	1185.00		NIST Webbook
ripol	1175.00		NIST Webbook
tb	439.00 ± 3.00	K	NIST Webbook
tc	611.91	K	Joback Method
tf	213.45	K	Joback Method
vc	0.483	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	301.90	J/molxK	611.91	Joback Method
cpg	292.23	J/molxK	580.80	Joback Method
cpg	282.08	J/molxK	549.70	Joback Method
cpg	271.43	J/molxK	518.59	Joback Method
cpg	260.26	J/molxK	487.49	Joback Method
cpg	248.56	J/molxK	456.38	Joback Method
cpg	236.30	J/molxK	425.28	Joback Method
dvisc	0.0064391	Paxs	213.45	Joback Method
dvisc	0.0003157	Paxs	425.28	Joback Method
dvisc	0.0004157	Paxs	389.98	Joback Method
dvisc	0.0005781	Paxs	354.67	Joback Method
dvisc	0.0008648	Paxs	319.37	Joback Method
dvisc	0.0014299	Paxs	284.06	Joback Method
dvisc	0.0027270	Paxs	248.75	Joback Method
hvapt	45.00	kJ/mol	386.50	NIST Webbook

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	338.20	K	2.80	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.54004e+01
Coeff. B	-4.05153e+03
Coeff. C	-6.32340e+01
Temperature range (K), min.	320.15
Temperature range (K), max.	464.82

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C1974045&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cp_g:	Ideal gas heat capacity
d_{visc}:	Dynamic viscosity
g_f:	Standard Gibbs free energy of formation
h_f:	Enthalpy of formation at standard conditions
h_{fus}:	Enthalpy of fusion at standard conditions
h_{vap}:	Enthalpy of vaporization at standard conditions
h_{vapt}:	Enthalpy of vaporization at a given temperature
log_{10ws}:	Log ₁₀ of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
pc:	Critical Pressure
pv_{ap}:	Vapor pressure
ri_{npol}:	Non-polar retention indices
ri_{pol}:	Polar retention indices
tb:	Normal Boiling Point Temperature
tb_{rp}:	Boiling point at reduced pressure
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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