

# Glutaric acid, 3-methylbenzyl pentyl ester

<b>Inchi:</b>	InChI=1S/C18H26O4/c1-3-4-5-12-21-17(19)10-7-11-18(20)22-14-16-9-6-8-15(2)13-16/h
<b>InchiKey:</b>	LOMCSHATQNSABD-UHFFFAOYSA-N
<b>Formula:</b>	C18H26O4
<b>SMILES:</b>	CCCCOC(=O)CCCC(=O)OCc1cccc(C)c1
<b>Mol. weight [g/mol]:</b>	306.40

## Physical Properties

Property code	Value	Unit	Source
gf	-264.38	kJ/mol	Joback Method
hf	-679.39	kJ/mol	Joback Method
hfus	41.60	kJ/mol	Joback Method
hvap	76.91	kJ/mol	Joback Method
log10ws	-4.74		Crippen Method
logp	3.942		Crippen Method
mvol	255.600	ml/mol	McGowan Method
pc	1552.45	kPa	Joback Method
rinpol	2503.00		NIST Webbook
tb	795.48	K	Joback Method
tc	995.15	K	Joback Method
tf	475.88	K	Joback Method
vc	0.984	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	761.25	J/molxK	795.48	Joback Method
cpg	829.35	J/molxK	961.87	Joback Method
cpg	817.75	J/molxK	928.59	Joback Method
cpg	805.16	J/molxK	895.32	Joback Method
cpg	791.55	J/molxK	862.04	Joback Method
cpg	776.92	J/molxK	828.76	Joback Method
cpg	839.97	J/molxK	995.15	Joback Method
dvisc	0.0000720	Paxs	795.48	Joback Method
dvisc	0.0000921	Paxs	742.21	Joback Method

dvisc	0.0001223	Paxs	688.95	Joback Method
dvisc	0.0001703	Paxs	635.68	Joback Method
dvisc	0.0002520	Paxs	582.41	Joback Method
dvisc	0.0004036	Paxs	529.15	Joback Method
dvisc	0.0007181	Paxs	475.88	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U377649&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U377649&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/60-734-0/Glutaric-acid-3-methylbenzyl-pentyl-ester.pdf>

Generated by Cheméo on 2024-04-23 15:03:01.13608308 +0000 UTC m=+16173830.056660392.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.