

# 1-Hexadecanol, acetate

<b>Other names:</b>	1-Acetoxyhexadecane 1-Hexadecanol, 1-acetate Acelan A Acetic acid, hexadecyl ester Acrylated lanolin alcohol Cetyl acetate ENT 1025 Hexadecanyl acetate Hexadecyl acetate NSC 8492 Palmityl acetate n-Hexadecyl acetate n-Hexadecyl ethanoate
<b>Inchi:</b>	InChI=1S/C18H36O2/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-20-18(2)19/h3-17H2,1-2
<b>InchiKey:</b>	LSTDYDRCKUBPDI-UHFFFAOYSA-N
<b>Formula:</b>	C18H36O2
<b>SMILES:</b>	CCCCCCCCCCCCCCCCOC(C)=O
<b>Mol. weight [g/mol]:</b>	284.48
<b>CAS:</b>	629-70-9

## Physical Properties

Property code	Value	Unit	Source
gf	-133.24	kJ/mol	Joback Method
hf	-659.65	kJ/mol	Joback Method
hfus	45.16	kJ/mol	Joback Method
hvap	102.30	kJ/mol	NIST Webbook
log10ws	-6.22		Crippen Method
logp	6.031		Crippen Method
mcvol	271.920	ml/mol	McGowan Method
pc	1189.06	kPa	Joback Method
rinpol	1994.00		NIST Webbook
rinpol	2013.00		NIST Webbook
rinpol	1996.00		NIST Webbook
rinpol	1996.00		NIST Webbook
rinpol	2003.00		NIST Webbook
rinpol	1976.00		NIST Webbook
rinpol	1991.10		NIST Webbook

rinpol	1984.00		NIST Webbook
rinpol	2009.00		NIST Webbook
rinpol	2010.00		NIST Webbook
rinpol	2010.00		NIST Webbook
rinpol	2010.00		NIST Webbook
rinpol	2010.00		NIST Webbook
rinpol	2009.00		NIST Webbook
rinpol	2009.00		NIST Webbook
rinpol	2013.00		NIST Webbook
rinpol	2008.00		NIST Webbook
rinpol	2006.90		NIST Webbook
rinpol	2013.00		NIST Webbook
rinpol	1995.00		NIST Webbook
rinpol	1997.00		NIST Webbook
rinpol	2006.00		NIST Webbook
rinpol	2011.00		NIST Webbook
rinpol	2004.00		NIST Webbook
rinpol	2005.00		NIST Webbook
rinpol	1990.00		NIST Webbook
rinpol	2003.00		NIST Webbook
rinpol	2009.00		NIST Webbook
rinpol	2009.00		NIST Webbook
rinpol	2009.00		NIST Webbook
rinpol	2010.00		NIST Webbook
rinpol	2009.00		NIST Webbook
rinpol	2009.00		NIST Webbook
rinpol	2008.00		NIST Webbook
ripol	2309.00		NIST Webbook
ripol	2294.00		NIST Webbook
ripol	2312.00		NIST Webbook
ripol	2312.00		NIST Webbook
ripol	2309.00		NIST Webbook
tb	687.53	K	Joback Method
tc	855.83	K	Joback Method
tf	292.35 ± 2.00	K	NIST Webbook
tf	295.40 ± 1.50	K	NIST Webbook
tf	291.70 ± 4.00	K	NIST Webbook
vc	1.067	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	890.54	J/mol×K	855.83	Joback Method
cpg	789.92	J/mol×K	687.53	Joback Method
cpg	808.70	J/mol×K	715.58	Joback Method
cpg	826.65	J/mol×K	743.63	Joback Method
cpg	843.80	J/mol×K	771.68	Joback Method
cpg	860.14	J/mol×K	799.73	Joback Method
cpg	875.72	J/mol×K	827.78	Joback Method
dvisc	0.0000953	Paxs	687.53	Joback Method
dvisc	0.0020226	Paxs	364.78	Joback Method
dvisc	0.0008765	Paxs	418.57	Joback Method
dvisc	0.0004595	Paxs	472.36	Joback Method
dvisc	0.0002749	Paxs	526.15	Joback Method
dvisc	0.0001809	Paxs	579.95	Joback Method
dvisc	0.0001278	Paxs	633.74	Joback Method
hvapt	70.30	kJ/mol	450.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	2.24900e+01
Coeff. B	-8.50187e+03
Coeff. C	-1.24394e+02
Temperature range (K), min.	507.32
Temperature range (K), max.	619.31

## Sources

**The Yaws Handbook of Vapor Pressure:**  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**KDB:**

<https://www.cheric.org/files/research/kdb/mol/mol1134.mol>

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C629709&Units=SI>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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