

# Cholesta-3,5-diene

<b>Other names:</b>	«delta»(3,5)-Cholestadiene Cholesterilene Cholestadiene 3,5-Cholestadiene Cholest-3,5-diene D3,5-cholestadiene
<b>Inchi:</b>	InChI=1S/C27H44/c1-19(2)9-8-10-20(3)23-14-15-24-22-13-12-21-11-6-7-17-26(21,4)25(2)
<b>InchiKey:</b>	RLHIRZFWJBOHHD-UHFFFAOYSA-N
<b>Formula:</b>	C27H44
<b>SMILES:</b>	CC(C)CCCC(C)C1CCC2C3CC=C4C=CCCC4(C)C3CCC12C
<b>Mol. weight [g/mol]:</b>	368.64
<b>CAS:</b>	747-90-0

## Physical Properties

Property code	Value	Unit	Source
gf	377.97	kJ/mol	Joback Method
hf	-256.88	kJ/mol	Joback Method
hfus	32.28	kJ/mol	Joback Method
hvap	73.76	kJ/mol	Joback Method
log10ws	-8.48		Crippen Method
logp	8.194		Crippen Method
mvol	339.250	ml/mol	McGowan Method
pc	1062.40	kPa	Joback Method
rinpol	2880.00		NIST Webbook
rinpol	2880.00		NIST Webbook
tb	859.03	K	Joback Method
tc	1084.16	K	Joback Method
tf	471.57	K	Joback Method
vc	1.290	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1191.70	J/mol×K	859.03	Joback Method

cpg	1221.42	J/mol×K	896.55	Joback Method
cpg	1250.97	J/mol×K	934.07	Joback Method
cpg	1280.70	J/mol×K	971.60	Joback Method
cpg	1310.96	J/mol×K	1009.12	Joback Method
cpg	1342.08	J/mol×K	1046.64	Joback Method
cpg	1374.41	J/mol×K	1084.16	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C747900&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C747900&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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