

# methyl (S)-(-)-lactate

<b>Other names:</b>	(S)-(-)-methyl lactate (S)-methyl 2-hydroxypropanoate L-methyl 2-hydroxypropanoate L-methyl lactate Propanoic acid, 2-hydroxy-, methyl ester, (S)- lactic acid, methyl ester, L- propanoic acid, 2-hydroxy-, (S)-, methyl ester
<b>Inchi:</b>	InChI=1S/C4H8O3/c1-3(5)4(6)7-2/h3,5H,1-2H3/t3-/m0/s1
<b>InchiKey:</b>	LPEKGGXMPWTOCB-VKHMYYHEASA-N
<b>Formula:</b>	C4H8O3
<b>SMILES:</b>	COC(=O)C(C)O
<b>Mol. weight [g/mol]:</b>	104.10
<b>CAS:</b>	27871-49-4

## Physical Properties

Property code	Value	Unit	Source
gf	-390.38	kJ/mol	Joback Method
hf	-528.20	kJ/mol	Joback Method
hfus	9.47	kJ/mol	Joback Method
hvap	49.94	kJ/mol	Joback Method
log10ws	0.27		Crippen Method
logp	-0.460		Crippen Method
mcvol	80.530	ml/mol	McGowan Method
pc	4717.12	kPa	Joback Method
tb	417.00 ± 0.80	K	NIST Webbook
tb	417.70	K	NIST Webbook
tc	636.26	K	Joback Method
tf	252.82	K	Joback Method
vc	0.296	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	166.01	J/mol×K	458.95	Joback Method

cpg	197.46	J/molxK	606.71	Joback Method
cpg	191.62	J/molxK	577.16	Joback Method
cpg	185.54	J/molxK	547.60	Joback Method
cpg	179.25	J/molxK	518.05	Joback Method
cpg	172.74	J/molxK	488.50	Joback Method
cpg	203.08	J/molxK	636.26	Joback Method
dvisc	0.0002093	Paxs	458.95	Joback Method
dvisc	0.0003436	Paxs	424.59	Joback Method
dvisc	0.0006157	Paxs	390.24	Joback Method
dvisc	0.0012346	Paxs	355.88	Joback Method
dvisc	0.0028726	Paxs	321.53	Joback Method
dvisc	0.0081807	Paxs	287.18	Joback Method
dvisc	0.0309613	Paxs	252.82	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	331.20	K	2.50	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C27871494&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C27871494&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Solid-Liquid Equilibria of N-Methylephedrine Enantiomers and Their Mixtures in Three Chiral Solvents Distinguished by Chain Length:</b>	<a href="https://www.doi.org/10.1021/je1007839">https://www.doi.org/10.1021/je1007839</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>t<sub>brp</sub>:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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