

# Phenylpropiolamide

<b>Other names:</b>	3-Phenylpropynoic acid amide
<b>Inchi:</b>	InChI=1S/C9H7NO/c10-9(11)7-6-8-4-2-1-3-5-8/h1-5H,(H2,10,11)
<b>InchiKey:</b>	HJZZEVFDPBDIDQ-UHFFFAOYSA-N
<b>Formula:</b>	C9H7NO
<b>SMILES:</b>	NC(=O)C#Cc1ccccc1
<b>Mol. weight [g/mol]:</b>	145.16
<b>CAS:</b>	7223-30-5

## Physical Properties

Property code	Value	Unit	Source
chs	-4589.40	kJ/mol	NIST Webbook
gf	277.64	kJ/mol	Joback Method
hf	200.95	kJ/mol	Joback Method
hfl	47.40 ± 4.00	kJ/mol	NIST Webbook
hfs	-20.00	kJ/mol	NIST Webbook
hfus	23.03	kJ/mol	Joback Method
hvap	57.44	kJ/mol	Joback Method
log10ws	-1.80		Crippen Method
logp	0.523		Crippen Method
mcvol	116.860	ml/mol	McGowan Method
pc	4621.41	kPa	Joback Method
tb	567.40	K	Joback Method
tc	826.44	K	Joback Method
tf	456.90	K	Joback Method
vc	0.428	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	249.36	J/mol×K	567.40	Joback Method
cpg	260.79	J/mol×K	610.57	Joback Method
cpg	271.30	J/mol×K	653.75	Joback Method
cpg	280.97	J/mol×K	696.92	Joback Method
cpg	289.83	J/mol×K	740.09	Joback Method

cpg	297.94	J/mol×K	783.27	Joback Method
cpg	305.34	J/mol×K	826.44	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7223305&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7223305&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/61-086-9/Phenylpropiolamide.pdf>

Generated by Cheméo on 2024-04-25 16:39:05.078879869 +0000 UTC m=+16352393.999457190.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.