

# 4-(Trifluoromethyl)benzamide

<b>Other names:</b>	p-Trifluoromethylbenzamide N-(4-Trifluoromethyl)benzamide Benzamide, 4-(trifluoromethyl)- 4-CF <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> CONH <sub>2</sub>
<b>Inchi:</b>	InChI=1S/C <sub>8</sub> H <sub>6</sub> F <sub>3</sub> NO/c9-8(10,11)6-3-1-5(2-4-6)7(12)13/h1-4H,(H2,12,13)
<b>InchiKey:</b>	WEJHBEDHLLBJFW-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>8</sub> H <sub>6</sub> F <sub>3</sub> NO
<b>SMILES:</b>	NC(=O)c1ccc(C(F)(F)F)cc1
<b>Mol. weight [g/mol]:</b>	189.13
<b>CAS:</b>	1891-90-3

## Physical Properties

Property code	Value	Unit	Source
affp	862.80	kJ/mol	NIST Webbook
basg	831.80	kJ/mol	NIST Webbook
gf	-524.80	kJ/mol	Joback Method
hf	-659.26	kJ/mol	Joback Method
hfus	18.75	kJ/mol	Joback Method
hvap	49.98	kJ/mol	Joback Method
log10ws	-2.75		Crippen Method
logp	1.804		Crippen Method
mcvol	116.680	ml/mol	McGowan Method
pc	3602.88	kPa	Joback Method
tb	535.08	K	Joback Method
tc	748.18	K	Joback Method
tf	356.24	K	Joback Method
vc	0.454	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	271.67	J/mol×K	535.08	Joback Method
cpg	282.12	J/mol×K	570.60	Joback Method
cpg	291.77	J/mol×K	606.11	Joback Method

cpg	300.65	J/mol×K	641.63	Joback Method
cpg	308.81	J/mol×K	677.15	Joback Method
cpg	316.31	J/mol×K	712.67	Joback Method
cpg	323.18	J/mol×K	748.18	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1891903&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1891903&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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