

# 3-(2-Ethylhexyloxy)-propylamine

<b>Other names:</b>	1-Propanamine, 3-[(2-ethylhexyl)oxy]- Propylamine, 3-(2-ethylhexoxy)- Propylamine, 3-((2-ethylhexyl)oxy)- 2-Ethylhexyl 3-aminopropyl ether 3-(2-Ethylhexoxy)propylamine NSC 1078 3-(2-Ethylhexoxy)propan-1-amine
<b>Inchi:</b>	InChI=1S/C11H25NO/c1-3-5-7-11(4-2)10-13-9-6-8-12/h11H,3-10,12H2,1-2H3
<b>InchiKey:</b>	DVFGEIYOLIFSRX-UHFFFAOYSA-N
<b>Formula:</b>	C11H25NO
<b>SMILES:</b>	CCCCC(CC)COCCCN
<b>Mol. weight [g/mol]:</b>	187.32
<b>CAS:</b>	5397-31-9

## Physical Properties

Property code	Value	Unit	Source
gf	0.75	kJ/mol	Joback Method
hf	-374.08	kJ/mol	Joback Method
hfus	27.11	kJ/mol	Joback Method
hvap	52.74	kJ/mol	Joback Method
log10ws	-2.71		Crippen Method
logp	2.568		Crippen Method
mvol	181.700	ml/mol	McGowan Method
pc	2018.13	kPa	Joback Method
rinpol	1357.00		NIST Webbook
rinpol	1357.00		NIST Webbook
tb	545.59	K	Joback Method
tc	721.22	K	Joback Method
tf	304.22	K	Joback Method
vc	0.693	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	458.32	J/mol×K	545.59	Joback Method
cpg	474.50	J/mol×K	574.86	Joback Method
cpg	490.03	J/mol×K	604.13	Joback Method
cpg	504.92	J/mol×K	633.40	Joback Method
cpg	519.18	J/mol×K	662.67	Joback Method
cpg	532.82	J/mol×K	691.94	Joback Method
cpg	545.86	J/mol×K	721.22	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5397319&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5397319&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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