

# 2-Phenoxyethyl caprylate

<b>Inchi:</b>	InChI=1S/C16H24O3/c1-2-3-4-5-9-12-16(17)19-14-13-18-15-10-7-6-8-11-15/h6-8,10-11H
<b>InchiKey:</b>	FTLLYZOWBWEERE-UHFFFAOYSA-N
<b>Formula:</b>	C16H24O3
<b>SMILES:</b>	CCCCCCCC(=O)OCCOc1ccccc1
<b>Mol. weight [g/mol]:</b>	264.36

## Physical Properties

Property code	Value	Unit	Source
gf	-142.67	kJ/mol	Joback Method
hf	-514.06	kJ/mol	Joback Method
hfus	35.21	kJ/mol	Joback Method
hvap	65.05	kJ/mol	Joback Method
log10ws	-4.22		Crippen Method
logp	3.969		Crippen Method
mvol	225.850	ml/mol	McGowan Method
pc	1744.82	kPa	Joback Method
rinpol	1916.00		NIST Webbook
rinpol	1916.00		NIST Webbook
tb	690.87	K	Joback Method
tc	885.06	K	Joback Method
tf	390.89	K	Joback Method
vc	0.866	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	628.67	J/molxK	690.87	Joback Method
cpg	703.35	J/molxK	852.69	Joback Method
cpg	690.23	J/molxK	820.33	Joback Method
cpg	676.22	J/molxK	787.96	Joback Method
cpg	661.30	J/molxK	755.60	Joback Method
cpg	645.46	J/molxK	723.23	Joback Method
cpg	715.59	J/molxK	885.06	Joback Method
dvisc	0.0000944	Paxs	690.87	Joback Method

dvisc	0.0001230	Paxs	640.87	Joback Method
dvisc	0.0001676	Paxs	590.88	Joback Method
dvisc	0.0002417	Paxs	540.88	Joback Method
dvisc	0.0003755	Paxs	490.88	Joback Method
dvisc	0.0006449	Paxs	440.89	Joback Method
dvisc	0.0012718	Paxs	390.89	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R540811&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R540811&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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