

# Perfluorobicyclo-[4.4.0]-dec-1,6-ene

<b>Inchi:</b>	InChI=1S/C10F16/c11-3(12)1-2(5(15,16)9(23,24)7(3,19)20)6(17,18)10(25,26)8(21,22)4(
<b>InchiKey:</b>	LSBZKGFVMTVTCT-UHFFFAOYSA-N
<b>Formula:</b>	C10F16
<b>SMILES:</b>	FC1(F)C2=C(C(F)(F)C(F)(F)C1(F)F)C(F)(F)C(F)(F)C(F)(F)C2(F)F
<b>Mol. weight [g/mol]:</b>	424.08
<b>CAS:</b>	54939-04-7

## Physical Properties

Property code	Value	Unit	Source
chl	-3550.20 ± 4.60	kJ/mol	NIST Webbook
gf	-3090.02	kJ/mol	Joback Method
hf	-3203.60 ± 4.80	kJ/mol	NIST Webbook
hfl	-3248.70 ± 4.80	kJ/mol	NIST Webbook
hfus	15.29	kJ/mol	Joback Method
hvap	45.30 ± 0.10	kJ/mol	NIST Webbook
log10ws	-6.16		Crippen Method
logp	5.393		Crippen Method
mcvol	154.060	ml/mol	McGowan Method
pc	1780.34	kPa	Joback Method
sl	491.20	J/mol×K	NIST Webbook
tb	430.10	K	Joback Method
tc	567.49	K	Joback Method
tf	425.26	K	Joback Method
tt	264.09 ± 0.02	K	NIST Webbook
vc	0.730	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	463.68	J/mol×K	544.59	Joback Method
cpg	410.94	J/mol×K	430.10	Joback Method
cpg	424.27	J/mol×K	453.00	Joback Method
cpg	436.04	J/mol×K	475.90	Joback Method
cpg	446.41	J/mol×K	498.79	Joback Method

cpg	455.57	J/mol×K	521.69	Joback Method
cpg	470.94	J/mol×K	567.49	Joback Method
cpl	428.40	J/mol×K	298.15	NIST Webbook
hfust	10.47	kJ/mol	263.00	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C54939047&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C54939047&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>sl:</b>	Liquid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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