

# 1,4-Benzenediamine, N,N-dimethyl-

<b>Other names:</b>	p-Phenylenediamine, N,N-dimethyl- p-(Dimethylamino)aniline p-(Dimethylamino)phenylamine p-Amino-N,N-dimethylaniline p-Aminodimethylaniline C.I. 76075 Dimethyl-p-phenylenediamine DMPD N,N-Dimethyl-p-benzenediamine N,N-Dimethyl-p-phenylenediamine N,N-Dimethyl-1,4-benzenediamine N,N-Dimethyl-1,4-phenylenediamine 4-(Dimethylamino)aniline 4-Amino-N,N-dimethylaniline Dimethyl-paraphenylenediamine N,N-Dimethyl-p-fenylendiamin 1,4-Benzenediamine, N1,N1-dimethyl- 4-(Dimethylamino)benzenamine NSC 1493 p-Phenylenediamine, N,N'-dimethyl-
<b>Inchi:</b>	InChI=1S/C8H12N2/c1-10(2)8-5-3-7(9)4-6-8/h3-6H,9H2,1-2H3
<b>InchiKey:</b>	BZORFPDSXLZWJF-UHFFFAOYSA-N
<b>Formula:</b>	C8H12N2
<b>SMILES:</b>	CN(C)c1ccc(N)cc1
<b>Mol. weight [g/mol]:</b>	136.19
<b>CAS:</b>	99-98-9

## Physical Properties

Property code	Value	Unit	Source
affp	955.00	kJ/mol	NIST Webbook
basg	928.40	kJ/mol	NIST Webbook
gf	296.49	kJ/mol	Joback Method
hf	117.93	kJ/mol	Joback Method
hfus	18.35	kJ/mol	Joback Method
hvap	49.02	kJ/mol	Joback Method
ie	6.46	eV	NIST Webbook
ie	6.46 ± 0.02	eV	NIST Webbook

log10ws	-1.01		Crippen Method
logp	1.335		Crippen Method
mcvol	119.780	ml/mol	McGowan Method
pc	3834.03	kPa	Joback Method
rinpol	1377.00		NIST Webbook
tb	536.20	K	NIST Webbook
tb	535.45 ± 2.00	K	NIST Webbook
tc	720.04	K	Joback Method
tf	334.59	K	Joback Method
vc	0.422	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	260.81	J/mol×K	499.07	Joback Method
cpg	274.36	J/mol×K	535.90	Joback Method
cpg	287.05	J/mol×K	572.73	Joback Method
cpg	298.92	J/mol×K	609.56	Joback Method
cpg	310.01	J/mol×K	646.38	Joback Method
cpg	320.35	J/mol×K	683.21	Joback Method
cpg	329.99	J/mol×K	720.04	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C99989&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C99989&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>aff:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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