

# Butane, 2-bromo-2-methyl-

<b>Other names:</b>	2-Bromo-2-methylbutane 2-Methyl-2-bromobutane tert-Amyl bromide tert-Pentyl bromide
<b>Inchi:</b>	InChI=1S/C5H11Br/c1-4-5(2,3)6/h4H2,1-3H3
<b>InchiKey:</b>	JOUWCKCVTDSMHF-UHFFFAOYSA-N
<b>Formula:</b>	C5H11Br
<b>SMILES:</b>	CCC(C)(C)Br
<b>Mol. weight [g/mol]:</b>	151.04
<b>CAS:</b>	507-36-8

## Physical Properties

Property code	Value	Unit	Source
gf	8.38	kJ/mol	Joback Method
hf	-128.95	kJ/mol	Joback Method
hfus	6.58	kJ/mol	Joback Method
hvap	31.86	kJ/mol	Joback Method
log10ws	-2.46		Crippen Method
logp	2.570		Crippen Method
mcvol	98.810	ml/mol	McGowan Method
pc	3857.88	kPa	Joback Method
ripol	734.00		NIST Webbook
ripol	750.00		NIST Webbook
ripol	925.00		NIST Webbook
ripol	939.00		NIST Webbook
ripol	937.00		NIST Webbook
tb	380.00	K	NIST Webbook
tc	574.49	K	Joback Method
tf	208.33	K	Joback Method
vc	0.366	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	162.75	J/molxK	376.73	Joback Method
cpg	173.57	J/molxK	409.69	Joback Method
cpg	183.72	J/molxK	442.65	Joback Method
cpg	193.25	J/molxK	475.61	Joback Method
cpg	202.20	J/molxK	508.57	Joback Method
cpg	210.59	J/molxK	541.53	Joback Method
cpg	218.45	J/molxK	574.49	Joback Method
dvisc	0.0067392	Paxs	208.33	Joback Method
dvisc	0.0032006	Paxs	236.40	Joback Method
dvisc	0.0017803	Paxs	264.46	Joback Method
dvisc	0.0011082	Paxs	292.53	Joback Method
dvisc	0.0007496	Paxs	320.60	Joback Method
dvisc	0.0005399	Paxs	348.66	Joback Method
dvisc	0.0004084	Paxs	376.73	Joback Method
hvapt	36.40	kJ/mol	358.50	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	380.20	K	98.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.36146e+01
Coeff. B	-2.58477e+03
Coeff. C	-9.26850e+01
Temperature range (K), min.	286.63
Temperature range (K), max.	403.99

## Sources

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C507368&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C507368&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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