

# Naphthalene, decahydro-1,4a-dimethyl-7-(1-methylethyl)-, [1S-(1«alpha»,4a«alpha»,7«alpha»,8a«beta»)]-

Other names:

4«alpha»-H-Eudesmane  
Selinane

Inchi:	InChI=1S/C15H28/c1-11(2)13-7-9-15(4)8-5-6-12(3)14(15)10-13/h11-14H,5-10H2,1-4H3/t
InchiKey:	DYEQPYSFRWUNNV-LJISPDSOSA-N
Formula:	C15H28
SMILES:	CC(C)C1CCC2(C)CCCC(C)C2C1
Mol. weight [g/mol]:	208.38
CAS:	30824-81-8

## Physical Properties

Property code	Value	Unit	Source
gf	125.17	kJ/mol	Joback Method
hf	-262.69	kJ/mol	Joback Method
hfus	14.80	kJ/mol	Joback Method
hvap	47.34	kJ/mol	Joback Method
log10ws	-4.68		Crippen Method
logp	4.885		Crippen Method
mcvol	200.490	ml/mol	McGowan Method
pc	1872.41	kPa	Joback Method
rinpol	1464.00		NIST Webbook
rinpol	1476.00		NIST Webbook
rinpol	1464.00		NIST Webbook
rinpol	1540.00		NIST Webbook
rinpol	1510.00		NIST Webbook
tb	563.62	K	Joback Method
tc	781.30	K	Joback Method
tf	281.03	K	Joback Method
vc	0.748	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	539.93	J/molxK	563.62	Joback Method

cpg	566.03	J/mol×K	599.90	Joback Method
cpg	590.56	J/mol×K	636.18	Joback Method
cpg	613.67	J/mol×K	672.46	Joback Method
cpg	635.49	J/mol×K	708.74	Joback Method
cpg	656.17	J/mol×K	745.02	Joback Method
cpg	675.83	J/mol×K	781.30	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C30824818&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C30824818&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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