

# benzyl 6-O-(«alpha»-L-arabinofuranosyl)-«beta»-D-glucopyranoside, TFA

Other names:	Benzenemethanol, «alpha»-L-Arabinofuranosyl-«beta»-D-Glucopyranoside, TFA
Inchi:	InChI=1S/C30H20F18O16/c31-25(32,33)19(49)57-8-11-13(61-21(51)27(37,38)39)15(63-39)23(45,46)47(48)29(35,36)30(2,3)42(43,44)34(40,41)44
InchiKey:	RLAJQRZPNBQZPM-NLBOSZRJSA-N
Formula:	C30H20F18O16
SMILES:	O=C(OCC1OC(OCC2OC(OCc3ccccc3)C(OC(=O)C(F)(F)F)C(OC(=O)C(F)(F)F)C2OC(=O)C(F)(F)F)O
Mol. weight [g/mol]:	978.44

## Physical Properties

Property code	Value	Unit	Source
gf	-4954.14	kJ/mol	Joback Method
hf	-6033.30	kJ/mol	Joback Method
hfus	106.78	kJ/mol	Joback Method
hvap	129.47	kJ/mol	Joback Method
log10ws	-7.26		Crippen Method
logp	4.146		Crippen Method
mcpvol	488.060	ml/mol	McGowan Method
pc	599.26	kPa	Joback Method
rinpol	2150.00		NIST Webbook
rinpol	2165.00		NIST Webbook
rinpol	2175.00		NIST Webbook
tb	1438.58	K	Joback Method
tc	2096.23	K	Joback Method
tf	998.58	K	Joback Method
vc	1.954	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1698.38	J/molxK	1438.58	Joback Method
cpg	1645.88	J/molxK	1548.19	Joback Method
cpg	1578.50	J/molxK	1657.80	Joback Method
cpg	1498.18	J/molxK	1767.41	Joback Method
cpg	1406.85	J/molxK	1877.02	Joback Method
cpg	1306.45	J/molxK	1986.63	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R394648&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R394648&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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