

# Benzenamine, 4-(octyloxy)-

<b>Other names:</b>	4-(Octyloxy)aniline p-Octyloxyaniline
<b>Inchi:</b>	InChI=1S/C14H23NO/c1-2-3-4-5-6-7-12-16-14-10-8-13(15)9-11-14/h8-11H,2-7,12,15H2,
<b>InchiKey:</b>	ACYGZCHBIGKPGR-UHFFFAOYSA-N
<b>Formula:</b>	C14H23NO
<b>SMILES:</b>	CCCCCCCCOc1ccc(N)cc1
<b>Mol. weight [g/mol]:</b>	221.34
<b>CAS:</b>	39905-45-8

## Physical Properties

Property code	Value	Unit	Source
gf	131.23	kJ/mol	Joback Method
hf	-205.66	kJ/mol	Joback Method
hfus	32.05	kJ/mol	Joback Method
hvap	62.75	kJ/mol	Joback Method
log10ws	-4.16		Crippen Method
logp	4.008		Crippen Method
mvol	200.210	ml/mol	McGowan Method
pc	2054.89	kPa	Joback Method
tb	646.33	K	Joback Method
tc	847.94	K	Joback Method
tf	391.97	K	Joback Method
vc	0.758	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	539.64	J/molxK	646.33	Joback Method
cpg	556.47	J/molxK	679.93	Joback Method
cpg	572.38	J/molxK	713.53	Joback Method
cpg	587.40	J/molxK	747.13	Joback Method
cpg	601.56	J/molxK	780.74	Joback Method
cpg	614.88	J/molxK	814.34	Joback Method
cpg	627.39	J/molxK	847.94	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	419.50 ± 1.50	K	0.07	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C39905458&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C39905458&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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