

4-Cyanobenzoic acid, 4-hexadecyl ester

Inchi: InChI=1S/C24H37NO2/c1-3-5-6-7-8-9-10-11-12-13-15-23(14-4-2)27-24(26)22-18-16-21(
InchiKey: KEOGVJFULGMUAY-UHFFFAOYSA-N
Formula: C24H37NO2
SMILES: CCCCCCCCCCCC(CCC)OC(=O)c1ccc(C#N)cc1
Mol. weight [g/mol]: 371.56

Physical Properties

Property code	Value	Unit	Source
gf	150.80	kJ/mol	Joback Method
hf	-398.83	kJ/mol	Joback Method
hfus	52.34	kJ/mol	Joback Method
hvap	91.20	kJ/mol	Joback Method
log10ws	-8.46		Crippen Method
logp	7.195		Crippen Method
mcvol	334.080	ml/mol	McGowan Method
pc	990.75	kPa	Joback Method
rinpola	2541.00		NIST Webbook
tb	958.11	K	Joback Method
tc	1174.00	K	Joback Method
tf	521.33	K	Joback Method
vc	1.315	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1101.58	J/molxK	958.11	Joback Method
cpg	1117.92	J/molxK	994.09	Joback Method
cpg	1133.02	J/molxK	1030.07	Joback Method
cpg	1146.94	J/molxK	1066.05	Joback Method
cpg	1159.74	J/molxK	1102.03	Joback Method
cpg	1171.46	J/molxK	1138.02	Joback Method
cpg	1182.18	J/molxK	1174.00	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U292454&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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