

# 1,1'-Biphenyl, 3,5-dichloro-4'-isopropyl

<b>Inchi:</b>	InChI=1S/C15H14Cl2/c1-10(2)11-3-5-12(6-4-11)13-7-14(16)9-15(17)8-13/h3-10H,1-2H3
<b>InchiKey:</b>	AXXUAYWMQQEWQC-UHFFFAOYSA-N
<b>Formula:</b>	C15H14Cl2
<b>SMILES:</b>	CC(C)c1ccc(-c2cc(Cl)cc(Cl)c2)cc1
<b>Mol. weight [g/mol]:</b>	265.18

## Physical Properties

Property code	Value	Unit	Source
gf	245.05	kJ/mol	Joback Method
hf	48.96	kJ/mol	Joback Method
hfus	26.39	kJ/mol	Joback Method
hvap	63.90	kJ/mol	Joback Method
log10ws	-6.63		Crippen Method
logp	5.784		Crippen Method
mcvol	199.170	ml/mol	McGowan Method
pc	2252.53	kPa	Joback Method
rinpol	2053.00		NIST Webbook
tb	685.32	K	Joback Method
tc	933.85	K	Joback Method
tf	394.05	K	Joback Method
vc	0.751	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	474.82	J/molxK	685.32	Joback Method
cpg	539.87	J/molxK	892.43	Joback Method
cpg	528.96	J/molxK	851.01	Joback Method
cpg	517.07	J/molxK	809.59	Joback Method
cpg	504.13	J/molxK	768.16	Joback Method
cpg	490.06	J/molxK	726.74	Joback Method
cpg	549.86	J/molxK	933.85	Joback Method
dvisc	0.0001347	Paxs	685.32	Joback Method
dvisc	0.0001686	Paxs	636.77	Joback Method

dvisc	0.0002189	Paxs	588.23	Joback Method
dvisc	0.0002980	Paxs	539.68	Joback Method
dvisc	0.0004311	Paxs	491.14	Joback Method
dvisc	0.0006763	Paxs	442.60	Joback Method
dvisc	0.0011855	Paxs	394.05	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R146320&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R146320&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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