

# Fumaric acid, decyl 2,4-dimethylpent-3-yl ester

<b>Inchi:</b>	InChI=1S/C21H38O4/c1-6-7-8-9-10-11-12-13-16-24-19(22)14-15-20(23)25-21(17(2)3)18
<b>InchiKey:</b>	VSCNDMTYLCLTFJ-CCEZHUSRSA-N
<b>Formula:</b>	C21H38O4
<b>SMILES:</b>	CCCCCCCCCOC(=O)C=CC(=O)OC(C(C)C)C(C)C
<b>Mol. weight [g/mol]:</b>	354.52

## Physical Properties

Property code	Value	Unit	Source
gf	-269.00	kJ/mol	Joback Method
hf	-864.99	kJ/mol	Joback Method
hfus	45.35	kJ/mol	Joback Method
hvap	79.45	kJ/mol	Joback Method
log10ws	-5.82		Crippen Method
logp	5.450		Crippen Method
mvol	317.330	ml/mol	McGowan Method
pc	1063.79	kPa	Joback Method
rinpol	2324.00		NIST Webbook
tb	835.30	K	Joback Method
tc	1026.77	K	Joback Method
tf	420.67	K	Joback Method
vc	1.222	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1005.90	J/molxK	835.30	Joback Method
cpg	1086.64	J/molxK	994.86	Joback Method
cpg	1072.59	J/molxK	962.95	Joback Method
cpg	1057.52	J/molxK	931.03	Joback Method
cpg	1041.40	J/molxK	899.12	Joback Method
cpg	1024.21	J/molxK	867.21	Joback Method
cpg	1099.70	J/molxK	1026.77	Joback Method
dvisc	0.0000309	Paxs	835.30	Joback Method
dvisc	0.0000433	Paxs	766.19	Joback Method

dvisc	0.0000648	Paxs	697.09	Joback Method
dvisc	0.0001059	Paxs	627.99	Joback Method
dvisc	0.0001954	Paxs	558.88	Joback Method
dvisc	0.0004290	Paxs	489.77	Joback Method
dvisc	0.0012192	Paxs	420.67	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348551&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348551&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/61-513-4/Fumaric-acid-decyl-2-4-dimethylpent-3-yl-ester.pdf>

Generated by Cheméo on 2024-04-28 03:56:25.39224192 +0000 UTC m=+16565834.312819236.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.