

DISOPYRAMIDE, M(N-DESALKYL-), AC

Inchi:	InChI=1S/C18H20N2O2/c1-14(21)20-13-12-18(17(19)22,15-8-4-2-5-9-15)16-10-6-3-7-11
InchiKey:	SUSVAGLOTXKSAV-UHFFFAOYSA-N
Formula:	C18H20N2O2
SMILES:	CC(=O)NCCC(C(N)=O)(c1cccc1)c1cccc1
Mol. weight [g/mol]:	296.36

Physical Properties

Property code	Value	Unit	Source
gf	226.34	kJ/mol	Joback Method
hf	-88.44	kJ/mol	Joback Method
hfus	36.54	kJ/mol	Joback Method
hvap	89.49	kJ/mol	Joback Method
log10ws	-3.51		Crippen Method
logp	1.984		Crippen Method
mcvol	240.060	ml/mol	McGowan Method
pc	2336.03	kPa	Joback Method
rinpol	2640.00		NIST Webbook
rinpol	2640.00		NIST Webbook
tb	891.81	K	Joback Method
tc	1138.77	K	Joback Method
tf	583.66	K	Joback Method
vc	0.892	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	731.36	J/mol×K	891.81	Joback Method
cpg	744.37	J/mol×K	932.97	Joback Method
cpg	756.29	J/mol×K	974.13	Joback Method
cpg	767.25	J/mol×K	1015.29	Joback Method
cpg	777.40	J/mol×K	1056.45	Joback Method
cpg	786.87	J/mol×K	1097.61	Joback Method
cpg	795.79	J/mol×K	1138.77	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R255088&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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