

# Glutaric acid, isobutyl pent-4-enyl ester

<b>Inchi:</b>	InChI=1S/C14H24O4/c1-4-5-6-10-17-13(15)8-7-9-14(16)18-11-12(2)3/h4,12H,1,5-11H2,2
<b>InchiKey:</b>	BUNWBVSNMTYPBP-UHFFFAOYSA-N
<b>Formula:</b>	C14H24O4
<b>SMILES:</b>	C=CCCCOC(=O)CCCC(=O)OCC(C)C
<b>Mol. weight [g/mol]:</b>	256.34

## Physical Properties

Property code	Value	Unit	Source
gf	-315.44	kJ/mol	Joback Method
hf	-701.74	kJ/mol	Joback Method
hfus	32.79	kJ/mol	Joback Method
hvap	64.01	kJ/mol	Joback Method
log10ws	-3.02		Crippen Method
logp	2.865		Crippen Method
mcvol	218.700	ml/mol	McGowan Method
pc	1707.53	kPa	Joback Method
rinpol	1754.00		NIST Webbook
rinpol	1754.00		NIST Webbook
tb	668.54	K	Joback Method
tc	849.17	K	Joback Method
tf	375.10	K	Joback Method
vc	0.843	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	599.41	J/molxK	668.54	Joback Method
cpg	668.83	J/molxK	819.06	Joback Method
cpg	656.40	J/molxK	788.96	Joback Method
cpg	643.26	J/molxK	758.85	Joback Method
cpg	629.38	J/molxK	728.75	Joback Method
cpg	614.77	J/molxK	698.64	Joback Method
cpg	680.53	J/molxK	849.17	Joback Method
dvisc	0.0001168	Paxs	668.54	Joback Method

dvisc	0.0001539	Paxs	619.63	Joback Method
dvisc	0.0002126	Paxs	570.73	Joback Method
dvisc	0.0003121	Paxs	521.82	Joback Method
dvisc	0.0004960	Paxs	472.91	Joback Method
dvisc	0.0008773	Paxs	424.01	Joback Method
dvisc	0.0018004	Paxs	375.10	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U359982&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U359982&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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