

# 1-Methyl-cis-2-(trans-3,4-methylenepentyl)-cyclopentane

<b>Inchi:</b>	InChI=1S/C10H18/c1-7-5-9(7)3-4-10-6-8(10)2/h7-10H,3-6H2,1-2H3/t7-,8+,9-,10-/m0/s1
<b>InchiKey:</b>	WFOQFWIOHOVKTQ-JXUBOQSCSA-N
<b>Formula:</b>	C10H18
<b>SMILES:</b>	CC1CC1CCC1CC1C
<b>Mol. weight [g/mol]:</b>	138.25

## Physical Properties

Property code	Value	Unit	Source
gf	139.40	kJ/mol	Joback Method
hf	-144.81	kJ/mol	Joback Method
hfus	20.07	kJ/mol	Joback Method
hvap	37.06	kJ/mol	Joback Method
log10ws	-2.83		Crippen Method
logp	3.079		Crippen Method
mcvol	130.040	ml/mol	McGowan Method
pc	2540.49	kPa	Joback Method
rinpola	966.20		NIST Webbook
rinpola	968.30		NIST Webbook
rinpola	968.90		NIST Webbook
rinpola	965.40		NIST Webbook
rinpola	968.30		NIST Webbook
tb	432.34	K	Joback Method
tc	621.26	K	Joback Method
tf	229.86	K	Joback Method
vc	0.507	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	286.50	J/mol×K	432.34	Joback Method
cpg	305.13	J/mol×K	463.83	Joback Method
cpg	322.77	J/mol×K	495.31	Joback Method
cpg	339.46	J/mol×K	526.80	Joback Method
cpg	355.24	J/mol×K	558.29	Joback Method

cpg	370.17	J/mol×K	589.77	Joback Method
cpg	384.30	J/mol×K	621.26	Joback Method
dvisc	0.0003717	Paxs	229.86	Joback Method
dvisc	0.0004332	Paxs	263.61	Joback Method
dvisc	0.0004877	Paxs	297.35	Joback Method
dvisc	0.0005359	Paxs	331.10	Joback Method
dvisc	0.0005787	Paxs	364.85	Joback Method
dvisc	0.0006169	Paxs	398.59	Joback Method
dvisc	0.0006510	Paxs	432.34	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R136784&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R136784&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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