

# Isobutyric acid, 2,3-dimethylphenyl ester

<b>Inchi:</b>	InChI=1S/C12H16O2/c1-8(2)12(13)14-11-7-5-6-9(3)10(11)4/h5-8H,1-4H3
<b>InchiKey:</b>	HXKJOHRJWKYBNW-UHFFFAOYSA-N
<b>Formula:</b>	C12H16O2
<b>SMILES:</b>	Cc1cccc(OC(=O)C(C)C)c1C
<b>Mol. weight [g/mol]:</b>	192.25

## Physical Properties

Property code	Value	Unit	Source
gf	-93.05	kJ/mol	Joback Method
hf	-327.50	kJ/mol	Joback Method
hfus	19.36	kJ/mol	Joback Method
hvap	54.67	kJ/mol	Joback Method
log10ws	-3.33		Crippen Method
logp	2.865		Crippen Method
mvol	163.620	ml/mol	McGowan Method
pc	2462.92	kPa	Joback Method
rinpol	1447.00		NIST Webbook
rinpol	1447.00		NIST Webbook
tb	586.45	K	Joback Method
tc	798.75	K	Joback Method
tf	333.62	K	Joback Method
vc	0.618	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	393.82	J/molxK	586.45	Joback Method
cpg	408.86	J/molxK	621.83	Joback Method
cpg	423.09	J/molxK	657.22	Joback Method
cpg	436.53	J/molxK	692.60	Joback Method
cpg	449.20	J/molxK	727.98	Joback Method
cpg	461.11	J/molxK	763.37	Joback Method
cpg	472.25	J/molxK	798.75	Joback Method
dvisc	0.0016439	Paxs	333.62	Joback Method

dvisc	0.0009045	Paxs	375.76	Joback Method
dvisc	0.0005614	Paxs	417.90	Joback Method
dvisc	0.0003803	Paxs	460.04	Joback Method
dvisc	0.0002750	Paxs	502.17	Joback Method
dvisc	0.0002091	Paxs	544.31	Joback Method
dvisc	0.0001653	Paxs	586.45	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U354638&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U354638&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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