

# 1,3-Dioxolane, 4-methyl-2-(2-methylpropyl), trans

<b>Other names:</b>	1,3-Dioxolane, 4-methyl-2-(2-methylpropyl), cis 2-isobutyl-4-methyl-1,3-dioxolane
<b>Inchi:</b>	InChI=1S/C8H16O2/c1-6(2)4-8-9-5-7(3)10-8/h6-8H,4-5H2,1-3H3
<b>InchiKey:</b>	PDVLTWPJDBXATJ-UHFFFAOYSA-N
<b>Formula:</b>	C8H16O2
<b>SMILES:</b>	CC(C)CC1OCC(C)O1
<b>Mol. weight [g/mol]:</b>	144.21
<b>CAS:</b>	18433-93-7

## Physical Properties

Property code	Value	Unit	Source
gf	-129.36	kJ/mol	Joback Method
hf	-437.59	kJ/mol	Joback Method
hfus	23.92	kJ/mol	Joback Method
hvap	41.98	kJ/mol	Joback Method
log10ws	-1.72		Crippen Method
logp	1.794		Crippen Method
mcvol	124.460	ml/mol	McGowan Method
pc	2925.00	kPa	Joback Method
rinpol	934.00		NIST Webbook
rinpol	932.00		NIST Webbook
rinpol	926.00		NIST Webbook
rinpol	939.80		NIST Webbook
rinpol	928.00		NIST Webbook
rinpol	941.00		NIST Webbook
rinpol	937.00		NIST Webbook
tb	446.51	K	Joback Method
tc	643.69	K	Joback Method
tf	224.72	K	Joback Method
vc	0.460	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	277.45	J/molxK	446.51	Joback Method
cpg	293.68	J/molxK	479.37	Joback Method
cpg	309.16	J/molxK	512.24	Joback Method
cpg	323.91	J/molxK	545.10	Joback Method
cpg	337.94	J/molxK	577.96	Joback Method
cpg	351.27	J/molxK	610.82	Joback Method
cpg	363.92	J/molxK	643.69	Joback Method
dvisc	0.0056311	Paxs	224.72	Joback Method
dvisc	0.0026179	Paxs	261.69	Joback Method
dvisc	0.0014712	Paxs	298.65	Joback Method
dvisc	0.0009386	Paxs	335.62	Joback Method
dvisc	0.0006547	Paxs	372.58	Joback Method
dvisc	0.0004874	Paxs	409.54	Joback Method
dvisc	0.0003810	Paxs	446.51	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C18433937&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C18433937&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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