

# Pentane, 2-chloro-3-methyl

<b>Inchi:</b>	InChI=1S/C6H13Cl/c1-4-5(2)6(3)7/h5-6H,4H2,1-3H3
<b>InchiKey:</b>	JPFZKJQZKBEPGQ-UHFFFAOYSA-N
<b>Formula:</b>	C6H13Cl
<b>SMILES:</b>	CCC(C)C(C)Cl
<b>Mol. weight [g/mol]:</b>	120.62
<b>CAS:</b>	24319-09-3

## Physical Properties

Property code	Value	Unit	Source
gf	-17.17	kJ/mol	Joback Method
hf	-193.47	kJ/mol	Joback Method
hfus	8.45	kJ/mol	Joback Method
hvap	32.56	kJ/mol	Joback Method
log10ws	-2.36		Crippen Method
logp	2.660		Crippen Method
mcvol	107.640	ml/mol	McGowan Method
pc	3035.62	kPa	Joback Method
rinpol	787.00		NIST Webbook
rinpol	787.00		NIST Webbook
tb	373.23	K	Joback Method
tc	553.82	K	Joback Method
tf	157.30	K	Joback Method
vc	0.408	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	185.75	J/mol×K	373.23	Joback Method
cpg	196.81	J/mol×K	403.33	Joback Method
cpg	207.42	J/mol×K	433.43	Joback Method
cpg	217.60	J/mol×K	463.52	Joback Method
cpg	227.35	J/mol×K	493.62	Joback Method
cpg	236.69	J/mol×K	523.72	Joback Method
cpg	245.63	J/mol×K	553.82	Joback Method

dvisc	0.0170744	Paxs	157.30	Joback Method
dvisc	0.0045226	Paxs	193.29	Joback Method
dvisc	0.0018178	Paxs	229.28	Joback Method
dvisc	0.0009357	Paxs	265.26	Joback Method
dvisc	0.0005644	Paxs	301.25	Joback Method
dvisc	0.0003793	Paxs	337.24	Joback Method
dvisc	0.0002752	Paxs	373.23	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C24319093&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C24319093&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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